

1 The importance of advection for CO₂ dynamics in the karst

2 Critical Zone: an approach from dimensional analysis

3 **Matthew D. Covington¹**

4 ¹*Department of Geosciences, University of Arkansas, Fayetteville, Arkansas, 72701, USA*

5 **ABSTRACT**

6 Karst landscapes provide unique challenges and opportunities to studies of processes
7 within the Critical Zone, which spans from the top of the canopy to the base of active
8 groundwater circulation. Dimensional analysis using the characteristic length scales and times
9 scales of karst processes enables development of an initial framework for quantification of the
10 rates and distribution of these processes throughout the Critical Zone. In particular, dimensional
11 analysis provides a useful tool to identify the relative importance of different processes and to
12 test the assumptions behind models of Critical Zone function. I briefly review prior use of
13 dimensional analysis to understand various aspects of the karst Critical Zone and then introduce
14 simple models for CO₂ transport within fractures and conduits of the vadose zone. Dimensional
15 analysis of these models suggests that advection through karst fractures within both the gas and
16 liquid phases will strongly impact vertical CO₂ profiles in the vadose zone in a variety of
17 settings. Implications of this finding for karst Critical Zone development and future data
18 campaigns are discussed.

19 **INTRODUCTION**

20 The Critical Zone (CZ) spans from the top of the canopy to the base of active
21 groundwater circulation (Brantley et al., 2006). Though an understanding of CZ processes is
22 crucial to predicting how Earth's surface will respond to climate change and anthropogenic

23 impacts, and to unraveling the connections between the physical, chemical, and biological
24 processes that drive landscape evolution, CZ processes and the coupling between them remain
25 relatively poorly understood. This has led to recent efforts toward a general and systematic
26 approach to CZ research as spearheaded by the Critical Zone Exploration Network (CZEN), a
27 network of field sites and community of researchers focused on understanding CZ processes.

28 Karst landscapes, which form in settings with highly soluble lithologies and occupy
29 roughly 12% of Earth's surface (Ford and Williams, 2007), provide an end-member case of CZ
30 dynamics. In contrast to other settings, karst systems evolve largely by congruent weathering.
31 Additionally, karst systems allow extremely rapid infiltration of water and the solutes and
32 particles transported by water. In karst settings, it is possible to have freshly recharged and
33 chemically aggressive water circulating at depths of hundreds or even thousands of meters within
34 minutes to hours after leaving the surface. One of the challenges of understanding CZ processes
35 is that they span many orders of magnitude in timescales (Brantley et al., 2006). This is
36 exacerbated in karst systems where hydraulic conductivities, and corresponding flow timescales,
37 span over 10 orders of magnitude depending upon the length scale over which they are measured
38 (Worthington, 2009). One advantage to karst CZ research is that, in many cases, the conduit flow
39 network that drains much of the CZ is physically accessible to humans in the form of caves.
40 Despite karst's unique position within the range of Earth's CZ dynamics, and relatively
41 widespread global distribution, a karst field site is not represented within the current CZEN, and
42 karst CZ dynamics are arguably even more poorly understood than other settings.

43 The preferential flow paths of dissolutionally enlarged fractures and conduits are of
44 central importance to understanding the dynamics of the karst CZ as they largely control the
45 distribution of water flow throughout the CZ and are the locations where geomorphic work is

46 focused. The distribution and apertures of such paths, and their ability to transmit fluids and
47 weathering products, will largely determine the nature of the karst landscape. Many processes
48 occurring within the CZ are controlled by gradients in temperature, solute concentration, carbon
49 dioxide, and the availability of dissolved and particulate organic matter. To understand the
50 variability of these quantities with depth, and as a function of flow path properties, the concept of
51 a process length scale is a useful tool. If a given process that occurs over a characteristic time
52 scale is coupled to flow along a karst fracture or conduit, then the combination of this
53 characteristic time scale with the flow velocity leads to a characteristic length scale (Covington
54 et al., 2012). These length scales provide a quantitative tool for dividing different dynamic
55 regimes, and for estimating the decay of critical zone processes with depth.

56 Process length scales are a specific example of dimensional analysis, whereby
57 characteristic length scales, time scales, and relevant dimensionless numbers are derived from
58 the equations governing a process. Here, I summarize previous use of such methods to
59 understand processes relevant to karst CZ dynamics and then use this framework to generate
60 questions and hypotheses concerning karst CO₂ dynamics and to test assumptions of current
61 models.

62 **DIMENSIONAL ANALYSIS OF PROCESSES IN THE KARST CRITICAL ZONE**

63 **Heat Transport**

64 Temperature is an important control on many of the chemical reactions that drive CZ
65 processes, including dissolution and microbial production of CO₂. Temperature variations at
66 Earth's surface propagate into the soil and rock and penetrate to a depth that is often referred to
67 as the thermal penetration depth, or skin depth (Incropera et al., 2007, pg. 299). This depth is
68 dependent on the time scale of temperature variations and is given by a relationship of the form

69 $\lambda_{skin} \sim \sqrt{D_{th} t},$ (1)

70 where D_{th} is the thermal diffusivity of rock, and t is the timescale of temperature variations.
71 Fluxes of air and water may enhance the depth of surface influence on soil and rock temperatures
72 within the subsurface. Specifically, karst conduits provide pathways that exchange both air and
73 water with the surface. Luetscher and Jeannin (2004) found that flow of both of these fluids can
74 result in substantial heat exchange within a karst system, and that air temperature with depth in a
75 karst aquifer normally trends with the adiabatic lapse rate, demonstrating the importance of air
76 circulation. Wigley and Brown (1971) derived characteristic length scales associated with
77 equilibration of atmospheric air to cave conditions. Using equations for the exchange of heat
78 between fluid and rock, one can derive the thermal length scale associated with fluid flow down
79 a fracture or conduit within the subsurface (Covington et al. 2012). All of these relationships
80 allow relatively simple calculations of the depth to which thermal variations should penetrate
81 into the karst CZ under different conditions. Roughly speaking, conductive heat changes
82 penetrate a few meters into the subsurface, air temperature changes penetrate a few hundreds of
83 meters, and water temperature changes can penetrate pathways multiple kilometers in length,
84 though relatively large (meter scale) conduits are needed for air and water temperature changes
85 to reach such distances into the subsurface.

86 **Dissolution**

87 The most widely discussed length scale in karst is the dissolutional penetration length
88 (Weyl, 1958; Dreybrodt, 1990; Palmer, 1991), the distance that undersaturated water will
89 penetrate along a flow path before approaching saturation. Within a linear kinetic regime, and at
90 a constant flow velocity, water will approach saturation exponentially over a length scale

91 $\lambda_d = \frac{V D_H}{4 \alpha_s},$ (2)

92 where V is the flow velocity, D_H is the hydraulic diameter of the flow path, and α_s is the linear
93 kinetic rate constant (for a derivation see Covington et al., 2012). However, calcite dissolution
94 kinetics become non-linear as saturation is approached, allowing slightly undersaturated water to
95 penetrate much deeper than it would otherwise (Dreybrodt, 1990; Palmer, 1991). Consideration
96 of the exponential length scale in Equation 2 allows an approximation of the distance over which
97 most of the aggressivity of the water is spent. Gabrovšek (2007; 2009) examined the vertical
98 distribution of dissolution rates as water percolates through the vadose zone. This analysis was
99 used to explore the relationship between measured solute fluxes and surface denudation rates
100 within a karst basin. Dissolution length scales have also been compared against CO₂ degassing
101 length scales and matrix inflow length scales to aid interpretation of observed downstream
102 decreases of dissolution rates in a cave stream (Covington et al., 2013). In general, dissolution
103 length scales are quite long (tens to hundreds of km) in mature karst conduits (Covington et al.,
104 2012), so changes in dissolution rates along a conduit are more likely to be a function of CO₂
105 dynamics or mixing of waters with different chemistries than dissolution occurring along the
106 flow path.

107 **Carbon Dioxide Dynamics**

108 Carbon dioxide, as the primary source of aggressivity in karst waters, provides an
109 important control on the distribution and rate of weathering processes within the karst Critical
110 Zone. CO₂ is produced within the soil zone due to root respiration and oxidation of organic
111 material. However, data from a number of sites also demonstrate that CO₂ concentrations can
112 increase with depth to values much higher than found in the soil zone, suggesting the presence of
113 deeper sources of CO₂ in some settings, likely as the result of organic matter decay deeper within
114 the vadose zone or at the water table (Atkinson, 1977; Wood and Petraitis, 1984; Baldini et al.,

115 2006; Whitaker and Smart, 2007a; Benavente et al., 2010). Models of vadose zone CO₂ profiles
116 within karst are typically based on diffusive transport equations (e.g. Wood, 1985; Breecker et
117 al., 2012), that assume that molecular diffusion is the primary mechanism for transport of CO₂
118 produced in the subsurface. Diffusive flux is described by Fick's 1st Law, which at steady state
119 gives

120 $F_{CO_2} = \frac{D_a}{L} [(CO_2)_L - (CO_2)_0], \quad (3)$

121 where F_{CO_2} is the flux of CO₂, D_a is the diffusion coefficient of CO₂ in air, L is the depth of the
122 CO₂ source, and (CO₂)_L and (CO₂)₀ are the concentrations of CO₂ at depth L and the surface,
123 respectively. An attractive feature of diffusive models is that they can explain relatively high
124 CO₂ concentrations at depth. At equilibrium, the flux of CO₂ out of the vadose zone must be
125 equal to the production rate within it. Therefore, even a modest production of CO₂ at depth will
126 result in an increase in CO₂ concentration at depth until a sufficient gradient is obtained to
127 sustain the required flux (Wood and Petraitis, 1984). However, the dissolutionally enlarged
128 pathways within a karst setting can allow for advective transport of species in both the liquid and
129 gaseous phases. A number of authors have noted that cave ventilation can be an important
130 control on CO₂ within a cave, and consequently on the timing and magnitude of calcite
131 precipitation on speleothems (Spötl et al., 2005; Wong et al., 2011; Frisia et al., 2011; Breecker
132 et al., 2012; Fairchild and Baker, 2012). However, less consideration has been given to whether
133 advective processes influence CO₂ dynamics within smaller aperture flow paths, such as
134 solutionally enlarged fractures. Below, I use dimensional analysis of simple mathematical
135 models of these processes to examine the potential influence of advection.

136 ***Advection of gaseous CO₂***

137 Since transport of CO₂ in the gas phase through the vadose zone is typically assumed to
138 be diffusive, yet karst vadose zones contain relatively open pathways that may allow advection
139 to occur, I will first examine the conditions under which one might expect for advection of gases
140 along fractures to be important. A number of mechanisms can drive cave airflow, although
141 arguably the most important type of cave airflow is chimney-effect airflow (Wigley and Brown,
142 1976), whereby temperature-related contrasts between surface and subsurface air density drive
143 buoyant flow between two or more entrances at different elevations. While this airflow
144 mechanism is well-documented within human-enterable caves, an interesting question is whether
145 it might drive significant flow within the fracture systems as well. The pressure difference, ΔP,
146 that drives chimney effect airflow can be approximated as

$$147 \quad \Delta P = \rho_{in} g h \frac{\Delta T}{T_{ext}}, \quad (4)$$

148 where ρ_{in} is the density of air inside the cave or fracture, g is Earth's gravitational acceleration, h
149 is the elevation difference between the two openings to the surface, ΔT is the temperature
150 difference between the subsurface and surface air, and T_{ext} is the surface air temperature in
151 Kelvin (e.g. Badino, 2010). Considering laminar flow within a fracture, the flow velocity is
152 given by the Hagen-Poiseuille equation,

$$153 \quad V = \frac{\Delta P a^2}{12 \mu_a L}, \quad (5)$$

154 Where V is the flow velocity, a is the fracture aperture, μ_a is the dynamic viscosity of air, and L
155 is the length of the flow path. For turbulent flow conditions, the flow velocity is given by the
156 Darcy-Weisbach equation

$$157 \quad V = \sqrt{\frac{4a\Delta P}{fL}}, \quad (6)$$

158 where f is the Darcy-Weisbach friction factor. The Peclet Number (Pe), which is the ratio of
159 diffusive and advective time scales, is used to quantify the relative importance of diffusion and
160 advection,

161
$$Pe = \frac{VL}{D}, \quad (7)$$

162 where D is the diffusion coefficient for the species of interest and L is the length scale of interest,
163 which is chosen here to be equal to the flow path length (Incropera et al., 2007). When $Pe \gtrsim 1$
164 advective transport is important, and when $Pe \gg 1$ then transport is dominated by advection
165 rather than diffusion (Fetter, 1993). Note that, unless flow velocity is zero, there will always
166 exist a length scale, L , above which advection is dominant. For laminar flow, Equations 4, 5, and
167 7 can be combined to give

168
$$Pe_{a,\text{lam}} = \frac{\rho_{\text{ing}} g h a^2}{12 \mu_a D_a} \frac{\Delta T}{T_{ext}}. \quad (8)$$

169 If one uses the flow path length, L , as the characteristic scale for which Pe is calculated, then for
170 the laminar flow case Pe is independent of flow path length. For turbulent flow, Equations 4, 6,
171 and 7 are combined giving

172
$$Pe_{a,\text{turb}} = \frac{L^{1/2}}{D_a} \sqrt{\frac{4\rho_{\text{ing}} g h a}{f} \frac{\Delta T}{T_{ext}}}. \quad (9)$$

173 For turbulent flow, Pe does depend on flow path length, and therefore calculation of Pe requires
174 more information about the geometry of the flow path than in the laminar case.

175 To quantify the potential importance of advective processes within fractures in the karst
176 vadose zone, I calculate Pe for a range of fracture apertures, a , and for entrance elevation
177 differences of 0.3, 3, 30, and 300 m (Figure 1). A temperature ratio (in Kelvin) of $\Delta T/T_{ext} =$
178 3.5% is assumed, which corresponds roughly to a difference between surface and subsurface
179 temperatures of 10°C with a subsurface temperature of 10°C. This choice is well within the

180 expected range for natural systems. Air properties are also taken at 10°C with $\rho_{in} = 1.25 \text{ kg m}^{-3}$,
181 $\mu_a = 1.8 \times 10^{-5} (\text{N}\cdot\text{s}) \text{ m}^{-2}$, and $D_a = 1.55 \times 10^{-5} \text{ m}^2\text{s}^{-1}$. For turbulent flow cases, a value of
182 $f = 0.1$ is used, though the conclusions do not depend strongly on this choice since all of the
183 turbulent flow cases are well within the advective regime. For turbulent flow, the flow path
184 length influences Pe, and therefore two example cases are chosen. In the first case (Figure
185 1A,B), L is set equal to the difference between the two entrance elevations, h . This is a limiting
186 case, as L cannot be any shorter. However, it provides a reasonable approximation to a vertical
187 fracture that intersects a horizontal cave (Figure 1A). For the second case $L=100 h$, which
188 represents a situation where the flow path length is much greater than the elevation difference
189 between the two fracture outlets (Figure 1C,D).

190 Figure 1 suggests that advection should be important in a wide variety of cases. For
191 example, even with a small elevation difference of 0.3 m, advection is an important process
192 within fractures with apertures on the order of 0.1 mm and larger. Differences between the short
193 (Figure 1A,B) and long flowpath (Figure 1C,D) cases are relatively minor because of: 1) the
194 independence of the laminar Pe on flow path length, and 2) the weak dependence of Pe on flow
195 path length in the turbulent regime. For the long flow path case, the transition to turbulence
196 occurs at slightly larger apertures and Pe, and the turbulent flow Pe values are also a factor of 10
197 larger. This results because of the positive effect of length on Pe that is only partially offset by
198 additional flow resistance.

199 The above analysis considers fractures that are open to the surface. While exposed
200 fractures are likely to occur in many karst settings, often the rock will be covered by a layer of
201 soil or regolith. Such a cover can substantially influence the extent of air exchange through the

202 fracture network below. If the cover behaves as a porous medium, then the specific discharge of
203 air through the cover is given by Darcy's Law,

204
$$q = \frac{\kappa \Delta P_p}{\mu_a L_p}, \quad (10)$$

205 where q is the specific discharge, κ is the permeability, ΔP_p is the pressure drop across the
206 porous cover, and L_p is the thickness of the cover. If the porous cover and fracture are connected
207 in series, then one can assume that the flow velocity in the fracture is equal to the specific
208 discharge in the cover and that the total chimney effect pressure drop (Equation 4) is equal to the
209 sum of the pressure drops across the cover and fracture. Using these assumptions, assuming
210 laminar flow, and again setting the characteristic length scale to the fracture flow path length,
211 leads to a new relation for Pe inside the fracture in the case of cover,

212
$$Pe_{a,cover} = \frac{\rho_{in}gh}{\mu_a D_a} \frac{\Delta T}{T_{ext}} \left(\frac{1}{12/a^2 + F/\kappa} \right), \quad (11)$$

213 where F is the ratio of cover thickness to fracture length. There are two limiting cases for
214 $Pe_{a,cover}$. When $\frac{12}{a^2} \gg \frac{F}{\kappa}$, then the bulk of the pressure loss is in the fracture, the cover has little
215 effect, and Equation 8 applies. Alternatively, when $\frac{12}{a^2} \ll \frac{F}{\kappa}$, then the bulk of the pressure loss is in
216 the cover, and the fracture properties no longer influence Pe. Comparing the magnitude of these
217 two terms allows estimation of the strength of the influence of the cover. In the limit where the
218 pressure loss is entirely in the cover, Pe approaches a maximum possible value of

219
$$Pe_{a,max} = \frac{\kappa \rho_{in}gh}{\mu_a D_a F} \frac{\Delta T}{T_{ext}}. \quad (12)$$

220 Two example cases are shown in Figure 2, where $F=0.05$ and κ is set to 10^{-9} m^2 in the first case
221 (Figure 2A) and 10^{-12} m^2 in the second (Figure 2B). One can see that there are realistic cases
222 where the cover would either allow advection to dominate or would shut it down. This model
223 likely produces an underestimate of the importance of advection, since it assumes that the air

224 exiting or entering the fracture will pass through a column of cover equal in cross sectional area
225 to the fracture. More likely, air would spread through a cone within the cover and perhaps
226 channel along preferential flow paths due to heterogeneity within the cover.

227 ***Advection of water and its influence on CO₂***

228 In the previous section I examined whether advection within the gaseous phase may play
229 an important role in the karst vadose zone. However, even in the absence of gas advection along
230 fractures and conduits, water is channeled along these flow paths, particularly during wet periods
231 or following recharge events. The water and air do not evolve in isolation but rather CO₂ will
232 exchange between the two. Here I will examine whether advective flows of water can influence
233 an otherwise diffusive CO₂ profile within a fracture. To examine this question, I use coupled
234 advection-dispersion equations for air and water within a fracture,

235
$$\frac{\partial(\text{CO}_2)_w}{\partial t} = D_w \frac{\partial^2(\text{CO}_2)_w}{\partial x^2} - V_w \frac{\partial(\text{CO}_2)_w}{\partial x} - \frac{D_w}{\delta_w^2} [(\text{CO}_2)_w - \alpha(\text{CO}_2)_a] + R, \quad (13)$$

236
$$\frac{\partial(\text{CO}_2)_a}{\partial t} = D_a \frac{\partial^2(\text{CO}_2)_a}{\partial x^2} - V_a \frac{\partial(\text{CO}_2)_a}{\partial x} + \frac{D_w}{\delta_w \delta_a} [(\text{CO}_2)_w - \alpha(\text{CO}_2)_a], \quad (14)$$

237 where t is time, x is position along the fracture, (CO₂)_a and (CO₂)_w are CO₂ concentrations in the
238 air and water, respectively, D_w is the diffusion coefficient for CO₂ in water, V_w and V_a are the
239 flow velocities of air and water, respectively, δ_w is the thickness of the water film on the fracture
240 wall, δ_a is the aperture of the air-filled portion of the fracture, α is the Ostwald solubility
241 coefficient for CO₂ (Wanninkhof et al., 2009), and R is a term representing CO₂ production
242 within the water or at the water-rock boundary. The geometry of this model is depicted in Figure
243 3. The exchange terms (the third terms on the right hand side of each equation) assume that CO₂
244 exchange is controlled by the rate of diffusion of CO₂ through the water film. This simple model
245 is arguably a gross oversimplification of dynamics in karst fractures, but dimensional analysis of
246 the model will allow an initial examination of assumptions and elucidation of relevant process

247 controls. One process not represented in Equations 13 and 14 is the interaction between CO₂
248 concentrations and dissolution or precipitation of calcite. These processes are discussed briefly
249 below, and can easily be introduced with an additional term, but are excluded here for the sake of
250 brevity.

251 To examine whether flowing water might influence CO₂ profiles within the vadose zone,
252 I consider a case where air velocity is zero, such that the CO₂ profile, if unperturbed by the flow
253 of water, would be controlled by diffusion. Setting the time derivative terms to zero allows us to
254 examine steady state CO₂ profiles, and for simplicity I also consider a case with no CO₂
255 production ($R=0$). To non-dimensionalize the model, I define a dimensionless position variable
256 $x^* = x/L$, and dimensionless concentration, $(CO_2^*)_i = (CO_2)_i / (CO_2)_0$, where L is a characteristic
257 length scale, and $(CO_2)_0$ is a reference concentration of CO₂. In the example solutions below,
258 $(CO_2)_0$ is set to the standard atmospheric concentration of CO₂, though any reference value could
259 be used. Using these definitions and assumptions, Equations 13 and 14 can be re-written as

260 $\frac{1}{Pe_w} \frac{\partial^2 (CO_2^*)_w}{\partial x^{*2}} = \frac{\partial (CO_2^*)_w}{\partial x^*} + \Lambda_w [(CO_2^*)_w - \alpha (CO_2^*)_a], \quad (15)$

261 $\frac{\partial^2 (CO_2^*)_a}{\partial x^{*2}} = -\xi [(CO_2^*)_w - \alpha (CO_2^*)_a], \quad (16)$

262 where $Pe_w = V_w L / D_w$, $\xi = D_w L^2 / D_a \delta_w \delta_a$, and $\Lambda_w = D_w L / \delta_w^2 V_w$ are all dimensionless numbers
263 that determine the nature of the solution. Pe_w is the Peclet Number for transport within the water
264 and determines the relative importance of advection and diffusion along the flowing film. $\xi =$
265 $(D_w L / \delta_w) / (D_a \delta_a / L)$ represents the ratio between the water-air CO₂ exchange rate and the
266 diffusive mass transport rate within the air along the length of the fracture. When ξ is large, then
267 exchange of CO₂ between the water and air will be important in comparison to diffusion in the
268 air along the fracture. When ξ is small then exchange is insignificant in comparison to diffusive
269 transport. The final dimensionless number, $\Lambda_w = L / \lambda_w$, is a ratio of the fracture length and the

length scale over which the water CO₂ concentration equilibrates with the air, $\lambda_w = V_w \delta_w^2 / D_w$. If Λ_w is small, then the water CO₂ concentration will be decoupled from the air. If Λ_w is large, then the CO₂ concentration in the water will remain nearly equilibrated to that of the air over length scales greater than L . These dimensionless numbers are summarized in Table 1.

(Table 1 near here)

In order for flow of water through the vadose zone to influence CO₂ profiles within the air, two physical conditions must hold: 1) the rate of transfer of CO₂ between air and water must be comparable to or greater than the rate of diffusive transport within the air, otherwise any perturbations to the profile via exchange will be efficiently smoothed by diffusion in the air, and 2) the advective mass transport rate of CO₂ in the water film must be comparable to or greater than the diffusive mass transport rate in the air, otherwise the water will not have the transport capacity to influence CO₂ concentrations in the air. The first condition can be quantified as $\xi \gtrsim 1$. The second condition is met when the ratio $\xi/\Lambda_w = \delta_w V_w / (\frac{D_a \delta_a}{L}) \gtrsim 1$, as this ratio expresses the ratio of the mass transport rate via advection in the water to the mass transport rate via diffusion in the air. For $\xi/\Lambda_w < 1$ the diffusive transport in the air outpaces advective transport in the water, and the film has no significant affect. This occurs in the limit of low film velocity and thin films.

To determine whether downward flow of water is likely to influence air CO₂ profiles in nature, we can examine typical values of the dimensionless numbers ξ and Λ_w under conditions that occur in natural systems. Concerning ξ , the ratio of the diffusion coefficients for CO₂ in air and water is roughly constant, $D_w/D_a \approx 10^{-4}$. Therefore we need only consider the other variables, L , which can be taken to be the thickness of the vadose zone or the length of the fracture flow path, and δ_a and δ_w , which are the aperture of the air-filled fracture and the

293 thickness of the water film, respectively. For a thin film of water flowing on a vertical wall, the
294 thickness of the film can be approximated as (Bird et al., 2002)

295
$$\delta_w = \sqrt{\frac{3V_w\mu_w}{g\rho_w}}. \quad (17)$$

296 Using this relationship and the ratio of diffusion coefficients above, the equations for ξ and Λ_w
297 can be rewritten in terms of V_w as

298
$$\xi = \frac{10^{-4}L^2}{\delta_a} \sqrt{\frac{g\rho_w}{3V_w\mu_w}} \quad (18)$$

299
$$\Lambda_w = \frac{LD_wg\rho_w}{3V_w^2\mu_w}. \quad (19)$$

300 To examine the likely ranges of these parameters in nature, I calculate ξ and Λ_w for
301 fractures/conduits with air-filled apertures of 0.001 m and 0.1 m and for vadose zone thicknesses
302 of $L = 0.3, 3, 30$, and 300 m. The dimensionless parameters are calculated as a function of V_w
303 over a range from 10^{-6} m s⁻¹ to 10 m s⁻¹ (Figure 4). ξ is small for only a few cases where the
304 vadose zone is very thin and the flow very fast. Therefore the ratio ξ/Λ_w , which expresses the
305 relative rates of mass transport in the water and air, is the primary determining factor for whether
306 the water influences the air CO₂ profile. The water advective influence is stronger for thicker
307 vadose zones and when the air-filled aperture is smaller (Figure 4A). However, in most settings
308 air CO₂ profiles are influenced by water advection for flow velocities above about 1 mm s⁻¹,
309 which is well within the range of expected velocities in natural fractures.

310 To illustrate the influence of these dimensionless numbers on the shape of equilibrium
311 CO₂ profiles, and the likely importance of advection, I numerically solve Equations 15 and 16 in
312 Python using the free and open source finite volume package FiPy (Guyer et al., 2009). The
313 equations are solved at steady-state for the boundary conditions

314
$$\frac{1}{\alpha}(\text{CO}_2^*)_w|_{x=0} = (\text{CO}_2^*)_a|_{x=0} = 1, \quad (20)$$

315 $(CO_2^*)_a|_{x=L} = 10$ (21)

316 $\frac{d(CO_2^*)_w}{dx}|_{x=L} = 0.$ (22)

317 The first two boundary conditions (20) set the air and water CO₂ at the top of the fracture to
318 values that are equilibrated with the atmosphere. The third condition (21) sets the CO₂
319 concentration at depth L equal to 10 times atmospheric, and the final condition (22) is an outlet
320 boundary condition at the downstream end of the water film. These boundary conditions
321 represent the case of a CO₂ source at depth L , which is only one of many possibilities. For
322 example, profiles would look quite different for a source near the surface. However, solutions for
323 this example set of boundary conditions will illustrate how the dimensionless parameters, and the
324 relative importance of physical processes that they represent, influence the form of the solution.
325 These conclusions also hold for other conditions, such as a near-surface CO₂ source, although the
326 shapes of the profiles will, in general, be different.

327 Solutions are shown for four example cases that span a variety of possible combinations
328 of the dimensionless parameters. Individual profiles are shown in Figure 5 and their location on
329 Figure 4 is also marked with identifying numbers. When ξ is small, the water film has no
330 significant influence on the air CO₂, and the profile closely matches that of the purely diffusive
331 case (Figure 5A). If $\xi > 1$ and $\Lambda_w < 1$ then the CO₂ air profile is strongly influenced by the water,
332 but the water film CO₂ concentration is not strongly affected by the air CO₂ as the water film is
333 moving too quickly to be significantly influenced (Figure 5B). When both ξ and Λ_w are
334 significantly larger than one and $\xi/\Lambda_w \gtrsim 1$, then the air CO₂ profile is significantly influenced
335 by the water film, but the water film stays near equilibrium with the air at all depths (Figure 5C).
336 Finally, when $\xi \gg 1$ and $\xi/\Lambda_w \lesssim 1$ then the CO₂ profile is nearly diffusive and the water profile

337 is close to equilibrium with the air, because the water film has insufficient transport capacity in
338 comparison to diffusion rates in the air (Figure 5D).

339 **DISCUSSION AND CONCLUSIONS**

340 While the models presented here are simple, and natural karst systems undoubtedly
341 display a broader and richer range of dynamics, these simple models provide an initial means of
342 thinking about the relative importance of different processes and the likely interactions between
343 them. In the past, vadose zone gas dynamics has typically been treated as diffusive, where
344 advective transport, within both the gas and liquid phases, is neglected (Penman, 1940a; Penman,
345 1940b; Marshall, 1959; Wood and Petraitis, 1984). Buoyancy driven flows have received recent
346 consideration within fracture systems (Weisbrod and Dragila, 2006; Weisbrod et al., 2009;
347 Nachshon et al., 2008). However, in karst systems dissolutional enlargement of fractures may
348 further increase the importance of advective processes.

349 In the models presented above, I first examine the possibility of advective air flows that
350 are driven by chimney effect pressure gradients. The influence of cave air flow patterns on CO₂
351 concentrations has been well-documented (Troester and White, 1984; Ek and Gewelt, 1985;
352 Spötl et al., 2005; Baldini et al., 2008; Milanolo and Gabrovšek, 2009; Wong et al., 2011;
353 Breecker et al., 2012; Gulley et al., 2014), and chimney effect airflow is arguably the most
354 common mechanism of cave airflow (Wigley and Brown, 1976). Previous authors have
355 demonstrated, using similar dimensional analysis, that advection should be a dominant process
356 for CO₂ transport in human-enterable caves (Frisia et al., 2011; Fairchild and Baker, 2012).
357 However, the dimensional analysis above suggests that chimney effect airflow should also be
358 important in many fractures or small solutionally enlarged pathways. For example, a fracture
359 with a 1 mm aperture that has two entrances with an elevation difference of only 0.3 m has a

360 Peclet Number of more than 100 in the case of a surface temperature 10 degrees different from
361 the subsurface temperature. This indicates that transport should be predominantly advective in
362 that case. Advection becomes even stronger with larger elevation differences. The suggested
363 importance of advection through smaller diameter flow paths is consistent with field data. For
364 example, Obir Cave, with no known upper entrance, displays strong chimney effect airflow
365 (Spötl et al., 2005).

366 Since fractures that are open to the surface should experience significant airflow in a
367 wide variety of conditions (Figure 1), the presence of significant chimney effect air flow is likely
368 to be primarily a function of cover. If there is a thick, relatively impermeable, cover then this will
369 shut down chimney effect airflow within the fracture network below it. The extent to which a
370 cover impedes airflow can be approximated by calculating the theoretical maximum Peclet
371 Number (Equation 12) appropriate for the cover thickness and permeability, though this
372 approach is likely to somewhat overestimate the capability of the cover to halt air circulation.
373 The fracture network pattern, and particularly the extent of horizontal connectivity, is also likely
374 to be an important control. A fracture network with more extensive horizontal connectivity is
375 more likely to host connected pathways that intersect the surface at substantially different
376 elevations.

377 Even if advection dominates CO₂ transport within fractures and caves, one might ask
378 whether this influence exists only in isolated zones in the vicinity of these preferential flow
379 paths. It may be the case that some portions of the vadose zone are isolated from the effects of
380 advection. However, diffusion occurs both vertically and laterally. Therefore the influence of
381 advection at a particular point within the vadose zone is likely to depend on the ratio of the
382 distance between that point and the surface and that point and the nearest advection-dominated

383 path (Figure 6). If, at a given location, the nearest advection-dominated path is closer than the
384 surface, then advection is likely to have an important influence at that location. Of course, any
385 anisotropy between vertical and lateral diffusion rates must also be accounted for.

386 Chimney effect airflow implies that there are regions of both inflow and outflow of air
387 that vary with season and surface temperature. Therefore, CO₂ dynamics can be expected to vary
388 diurnally and seasonally. Lower and upper entrances will also experience substantially different
389 CO₂ dynamics and gradients with depth, since each undergoes the opposite direction of flow at
390 any given time. Such effects may have important implications for the temporal and spatial
391 distribution of geomorphic work within the karst vadose zone. In comparison to chimney effect
392 airflow, karst systems dominated by barometric air exchange would be expected to display quite
393 different distributions of CO₂ in space and time.

394 The model presented for the coupling between water and air CO₂ concentrations suggests
395 that flowing water, even in the absence of air flow, is likely to alter CO₂ profiles away from
396 purely diffusive profiles to curved profiles that display lower gradients near the surface and
397 higher gradients at depth. The exact shape will depend on the boundary conditions at the water
398 table and the distribution of CO₂ production with depth. In particular, the example cases shown
399 in Figure 5 assume CO₂ production is located at depth. Different distributions of CO₂ production
400 can result in quite different vertical CO₂ profiles than the ones depicted here. The strength of the
401 advective influence of water depends on several parameters, but, for most of the parameter space
402 sampled, water flow velocities of 1 mm s⁻¹ and greater typically result in a CO₂ profile that is
403 modified by water advection. Within natural systems, such effects are likely to be event driven,
404 and the profile may evolve seasonally and with individual recharge events. The ability for a
405 diffusive profile to recover between recharge events will be largely dependent on the ratio of the

406 diffusion time scale for the vadose zone thickness to the time period between recharge events
407 that modify the profile.

408 There are a variety of processes that are not considered explicitly here, but for which a
409 similar approach may enable fruitful progress. One might ask whether there are typical length
410 scales that govern CO₂ production within the subsurface. For example, roots may penetrate to
411 typical depths that vary by plant species and other parameters. Filtration and processing of
412 particulate and dissolved organic matter may also have associated length scales that depend upon
413 rates of microbial decay, soil porosity and permeability, fracture aperture, recharge rate, and
414 temperature. Understanding these controls will substantially improve our ability to predict the
415 distribution of CO₂ within karst Critical Zones in different settings. Any length scales associated
416 with CO₂ production could be compared against length scales of water-air CO₂ exchange to
417 quantify the influence that production would have on the overall profile.

418 Similarly, comparing dissolutional length scales with CO₂ exchange and production length
419 scales would improve our understanding of the distribution of geomorphic work throughout the
420 karst Critical Zone. Under what conditions is aggressivity largely spent near the zone where it is
421 produced versus penetrating much greater depths before the CO₂ has reacted? In cases where
422 dissolutional length scales are much shorter than CO₂ exchange length scales then dissolution
423 and/or precipitation is also likely itself to be an important driver of CO₂ dynamics. Similar
424 coupling between wind-driven ventilation and dissolution/precipitation has been observed within
425 a semi-arid karst setting (Roland et al., 2013).

426 The analysis presented here, motivated from a physics-based approach, is entirely
427 theoretical. One might ask what implications it has for field studies of CO₂ dynamics in karst. I
428 would argue that there are several important implications. First, while prior models have

429 typically assumed CO₂ transport in the vadose zone is diffusive, relatively simple models of
430 buoyancy driven flow and transport within the karst vadose zone suggest that typical conditions
431 can produce advective effects driven by flow of air or water. Therefore, interpretation of vadose
432 zone CO₂ dynamics in karst should not assume from the start that diffusion is dominant and that
433 flow of water does not perturb CO₂ concentrations. The model presented here also provides a
434 mathematical starting point for quantifying advective effects and understanding the influence
435 they may have on CO₂ distributions in time and space. CO₂ profiles under the influence of
436 advection are more likely to contain sharp features that correlate with different air or water
437 circulation zones and to change dynamically with changes in surface temperature and recharge
438 rate. Such effects have been observed within a deep karst vadose zone in Spain within boreholes
439 where air flow was also observed (Benavente et al., 2010).

440 Spatial and temporal variability in CO₂ may also play an important role in
441 speleogenesis (Gabrovšek and Dreybrodt, 2000; Gabrovšek et al., 2000; Whitaker and Smart,
442 2007b; Gulley et al., 2013; Gulley et al., 2014), and advective processes are likely to be an
443 important control on the spatial and temporal distribution of dissolution within mature cave
444 systems. The distribution of CO₂ within the vadose zone, the rate of exchange of CO₂ between
445 air and water, and the interactions between water and rock via dissolution or precipitation all
446 influence carbon isotope fractionation (Fairchild and Baker, 2012, p. 176-180). Therefore, the
447 conclusions here concerning the importance of advection have important implications for studies
448 of paleoclimate that employ speleothems.

449 In summary, the models presented here suggest that CO₂ dynamics within the karst Critical
450 Zone may be controlled by a number of processes that have traditionally been neglected.
451 Currently, there is little data available to constrain the range of dynamics that occur in natural

452 systems. Data campaigns that sample densely in both space and time may provide substantial
453 insight into the processes that drive landform and subsurface evolution in karst terrains.

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455 I appreciate stimulating discussions with Matija Perne and Joe Myre during the development of
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457 Jeannin also helped to substantially clarify and sharpen this work.

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585 **TABLES**

586 Table 1. Dimensionless numbers that determine the dynamics of the coupled water-air CO₂
 587 transport model.

Dimensionless Number	Physical interpretation
$Pe_w = V_w L / D_w$	Relative importance of advective and dispersive mass transport within the water film. Advection dominates for large Pe_w .
$\xi = D_w L^2 / D_a \delta_w \delta_a$	Relative importance of mass transfer via water-air exchange and diffusion within the air. Exchange is important for large ξ .
$\Lambda_w = D_w L / \delta_w^2 V_w$	Ratio of fracture length to the length scale associated with CO ₂ exchange. For large Λ_w water and air CO ₂ are closely coupled.
$\frac{\xi}{\Lambda_w} = L \delta_w V_w / D_a \delta_a$	Relative importance of advective mass transfer in the water and diffusive mass transfer in the air. When this ratio is large mass transfer in the water is important.

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589 **Table 2.** Description of notation

Symbol	Definition
a	fracture aperture
α	Ostwald solubility coefficient
α_s	calcite dissolution kinetic rate constant
(CO ₂), (CO ₂) _a , (CO ₂) _w	concentration of CO ₂ , in air, in water
D_a	diffusion coefficient for CO ₂ in air
D_w	diffusion coefficient for CO ₂ in water
D_H	hydraulic diameter
D_{th}	thermal diffusivity of rock
δ_a	aperture of air-filled portion of fracture
δ_w	thickness of water film on fracture wall
f	Darcy-Weisbach friction factor
F	ratio of cover thickness to fracture length (L_p/L)
F_{co2}	flux rate of CO ₂
g	Earth's gravitational acceleration
h	difference in elevation between two entrances
κ	permeability
L	fracture length or characteristic length scale
L_p	thickness of porous cover
Λ_w	$D_w L / \delta_w^2 V_w$, dimensionless number (see Table 1)
λ_d	dissolutional penetration length
λ_w	CO ₂ exchange length scale

λ_{skin}	thermal skin depth
μ_a	dynamic viscosity of air
ΔP	pressure difference
ΔP_p	pressure difference across the porous cover
Pe	Peclet Number
ρ_{in}	density of air inside the cave or fracture
q	specific discharge (of air)
R	CO ₂ production rate
ΔT	difference in temperature inside and outside cave or fracture
T_{ext}	temperature outside of cave or fracture
t	time
V (V_a or V_w)	velocity (of air or water)
x	position along the fracture
ξ	$D_w L^2 / D_a \delta_w \delta_a$, dimensionless number (see Table 1)

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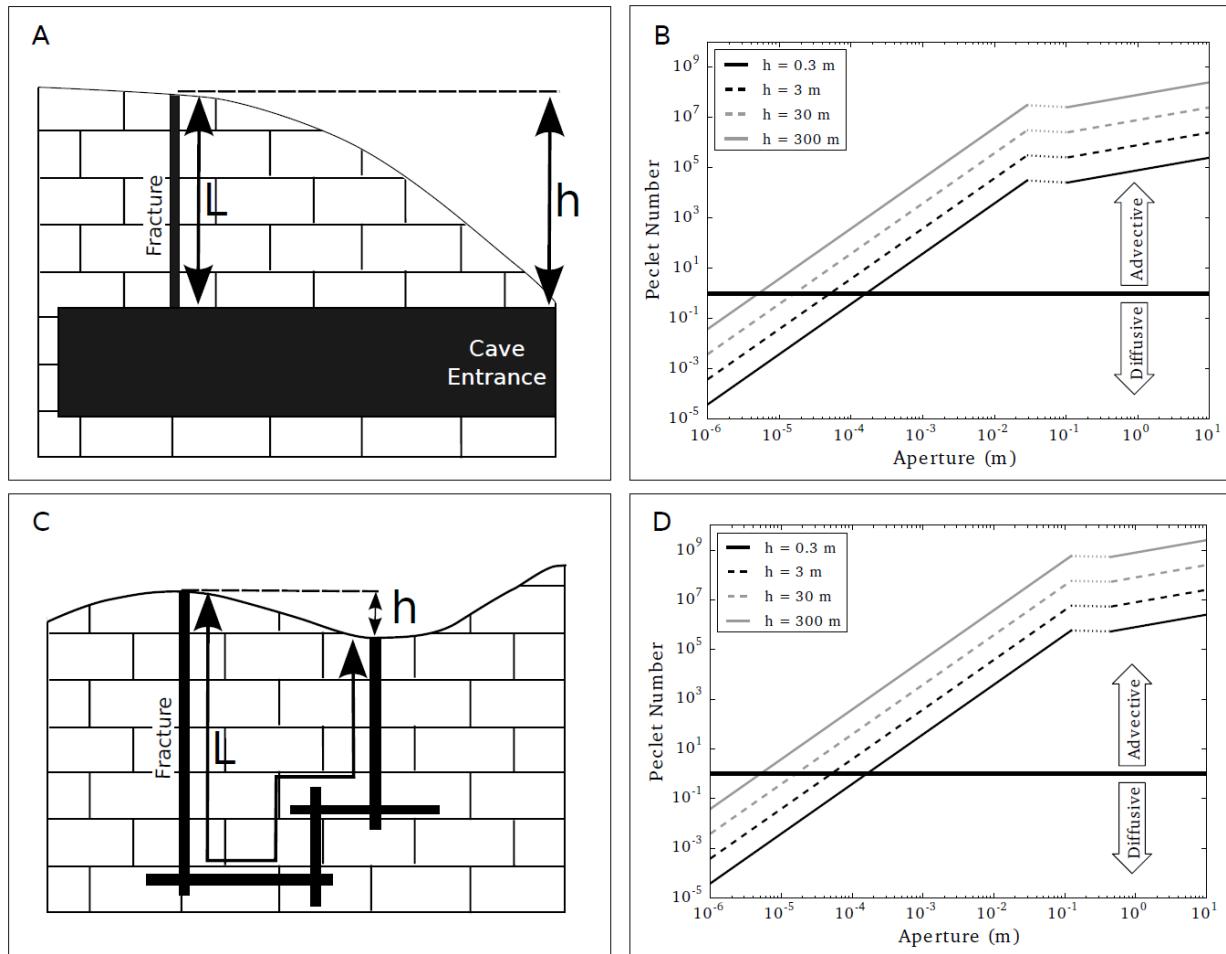
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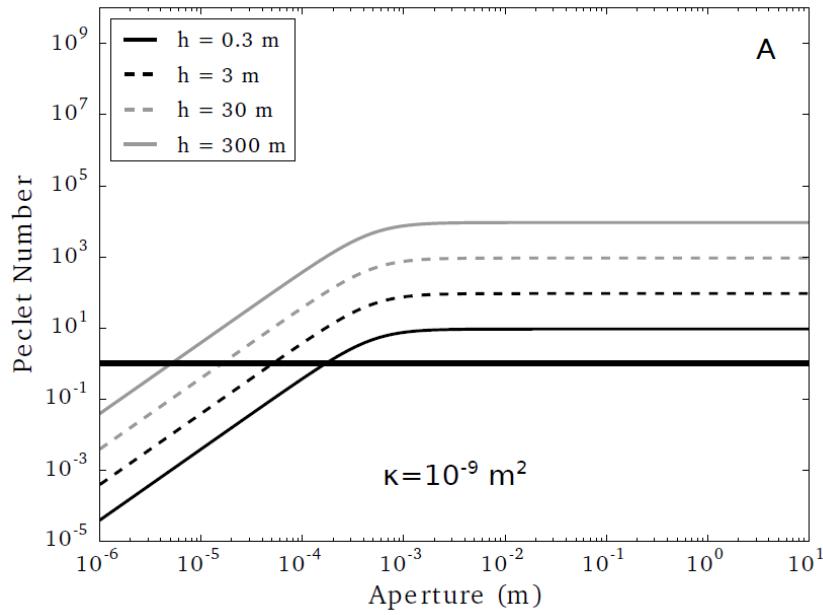
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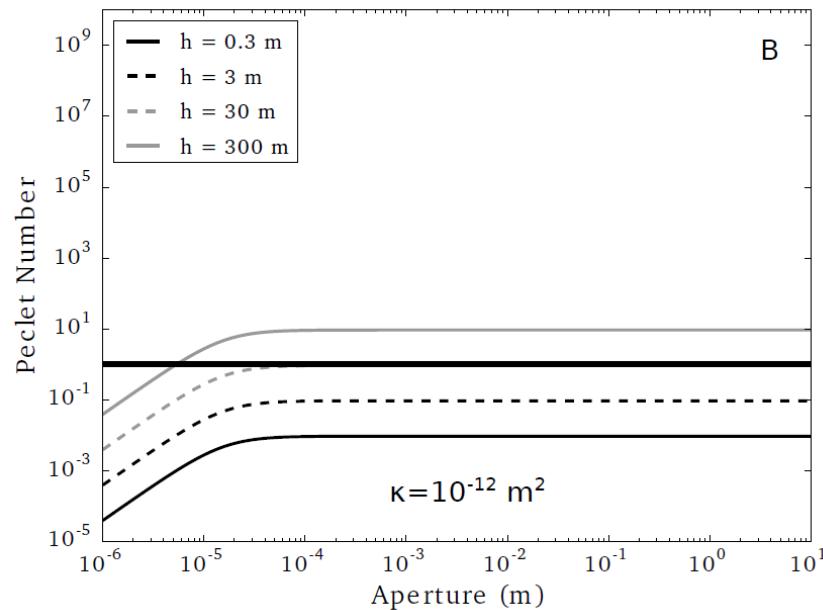
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608 Figure 1. The values of Peclet Number (Pe) for chimney effect airflow in fractures for a range of
 609 fracture apertures and differences between entrance elevations (h). Transport is dominated by
 610 advection for $Pe \gg 1$. Panel A shows an example geometry where the fracture length, L , is equal
 611 to the elevation difference between entrances and Pe is calculated for this case in Panel B. Panel
 612 C shows a case where fracture length is much greater than the elevation difference between
 613 entrances, and Pe is calculated for $L=100 h$ in Panel D. Note that this difference has no influence
 614 on Pe for the laminar flow regime (i.e. in the left portion of the diagrams). Laminar to turbulent
 615 transition occurs within the region with the dotted lines.

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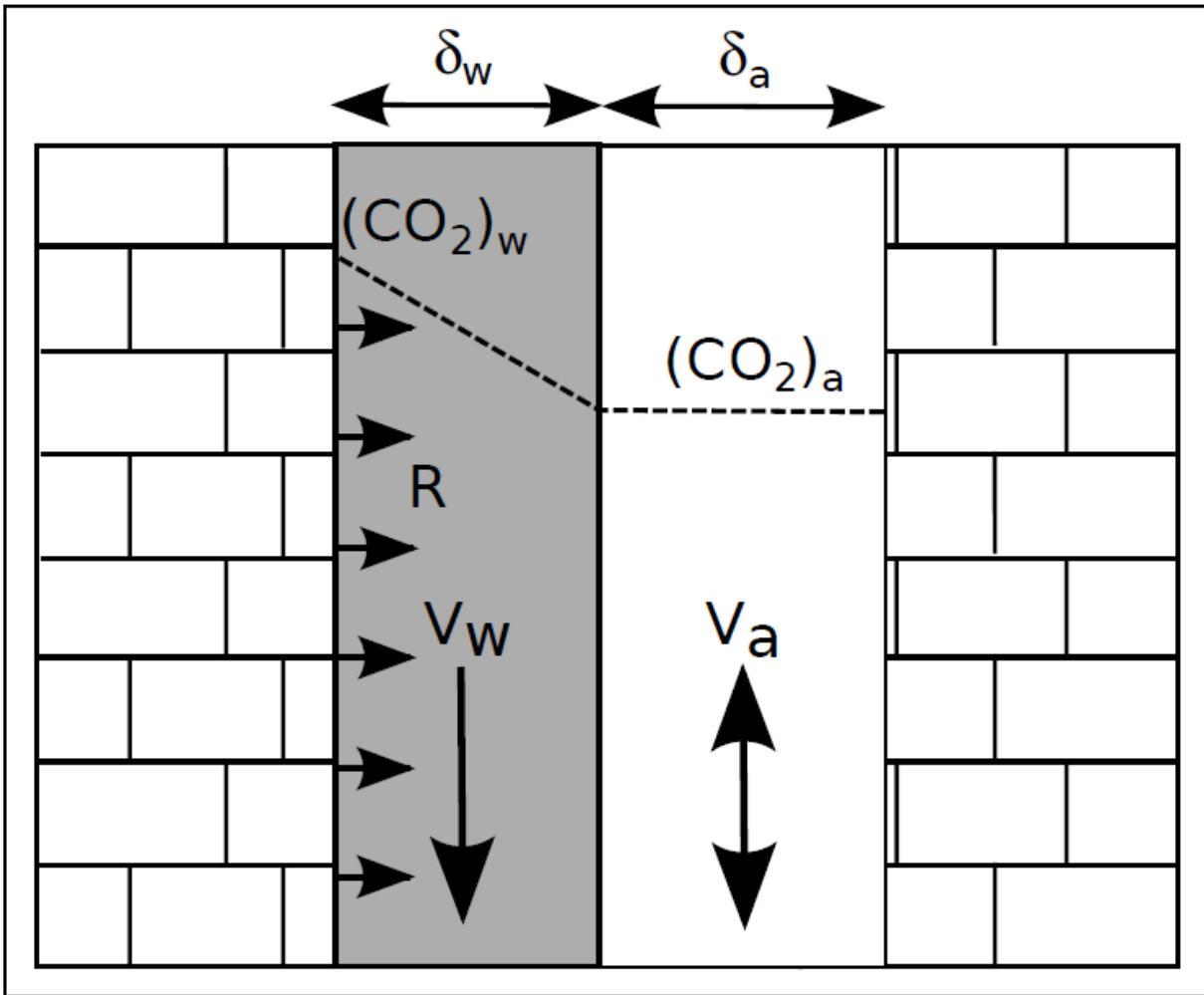


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619 Figure 2. Peclet Numbers (Pe) calculated for chimney effect airflow through fractures with a
 620 porous cover. In this case, a maximum value of Pe is reached for a sufficiently large fracture
 621 aperture, as all pressure loss occurs in the porous media in that limit. Panel A depicts a case
 622 where permeability $\kappa=10^{-9} \text{ m}^2$, and Panel B depicts a case where $\kappa=10^{-12} \text{ m}^2$. For both cases, the
 623 cover thickness is assumed to be 5% of the fracture length.

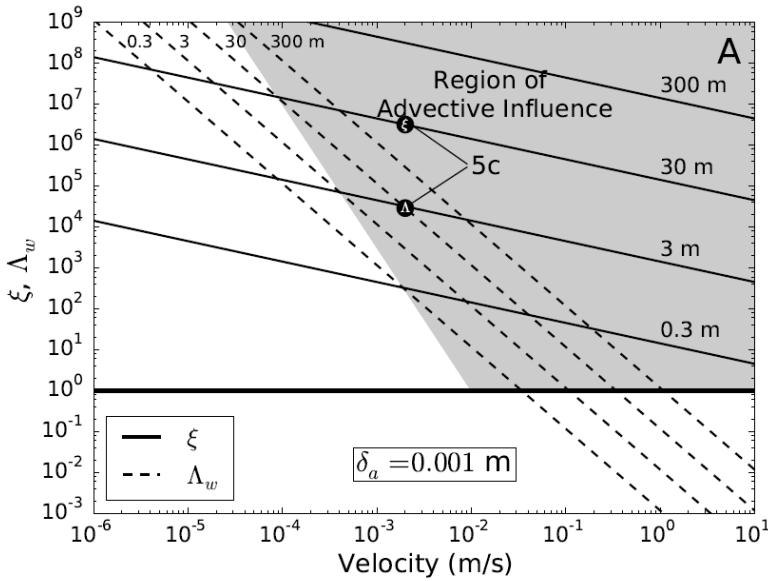


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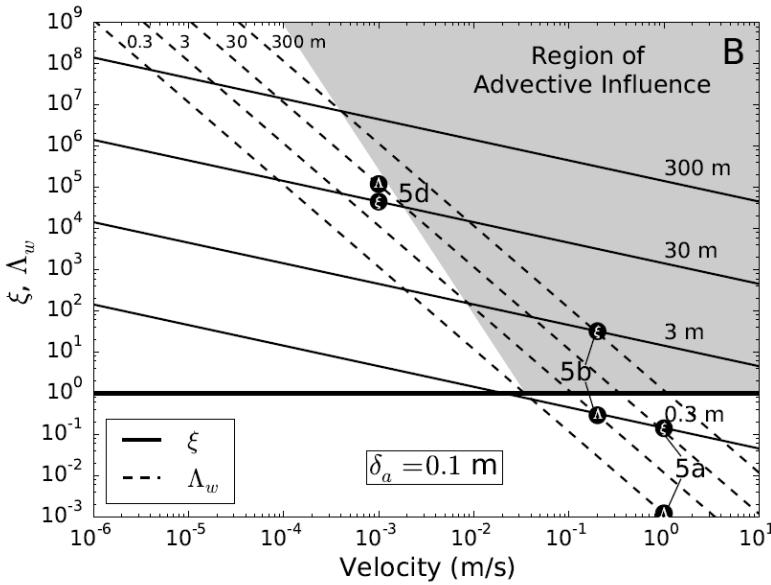
625 Figure 3. The geometry of the assumed model for transport of CO₂ within the air and water
 626 flowing in a fracture given by Equations 13 and 14. The dashed line depicts CO₂ concentration,
 627 which is assumed to be uniform across the air-filled cross-section and linear across the water
 628 film. Exchange is assumed to be controlled by the rate of diffusion within the water film. R
 629 depicts the production of CO₂ at the rock-water boundary.

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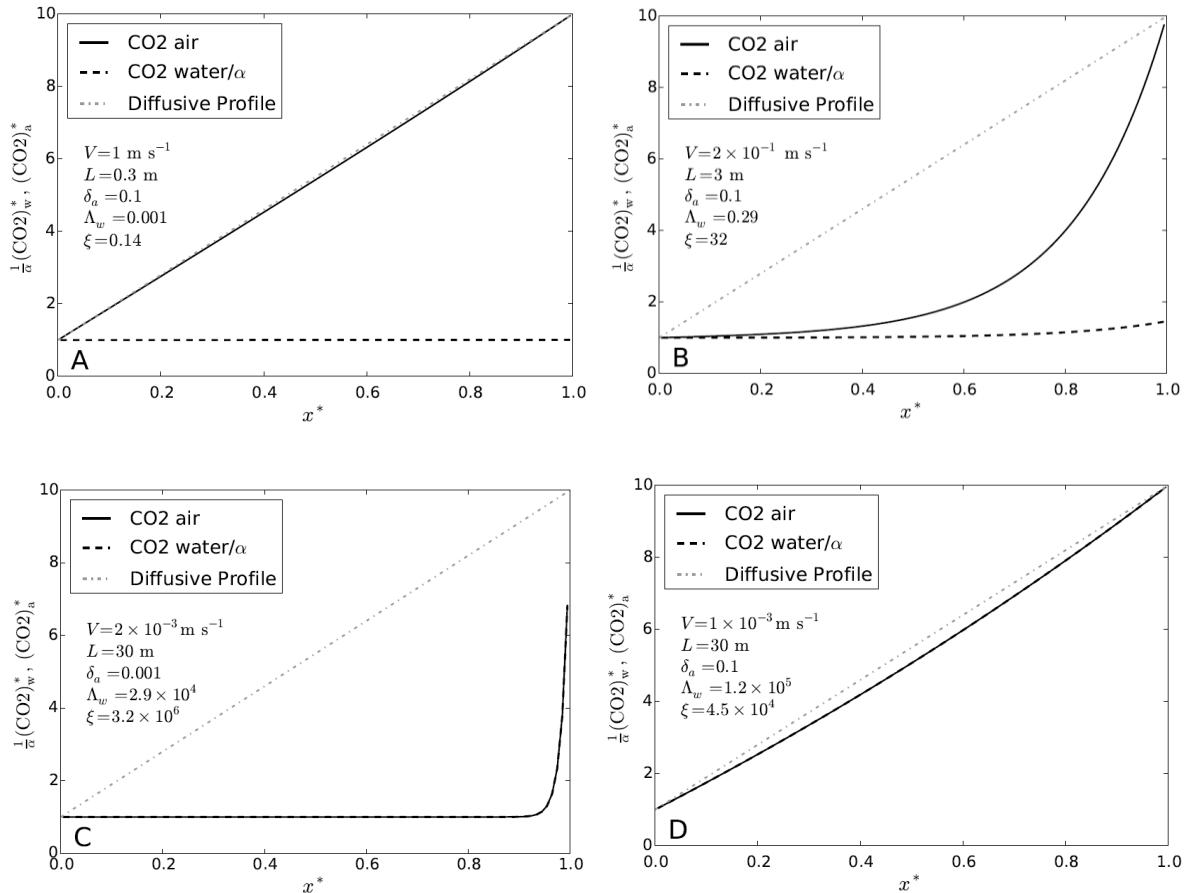


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634 Figure 4. The dimensionless numbers ξ (solid) and Λ_w (dashed) as a function of flow velocity for
 635 different values of fracture length ($L = 0.3 \text{ m}, 3 \text{ m}, 30 \text{ m}, \text{ and } 300 \text{ m}$). Advection is important in
 636 determining the CO₂ profile with depth when $\xi > 1$ and $\xi > \Lambda_w$ (shaded region). Panel (A) shows a
 637 case where the air-filled fracture aperture is 0.001 m , and Panel (B) shows a case where the air-
 638 filled fracture aperture is 0.1 m . The letters and numbers (5a-d) label values of ξ and Λ_w for the
 639 example profiles shown in Figure 5.



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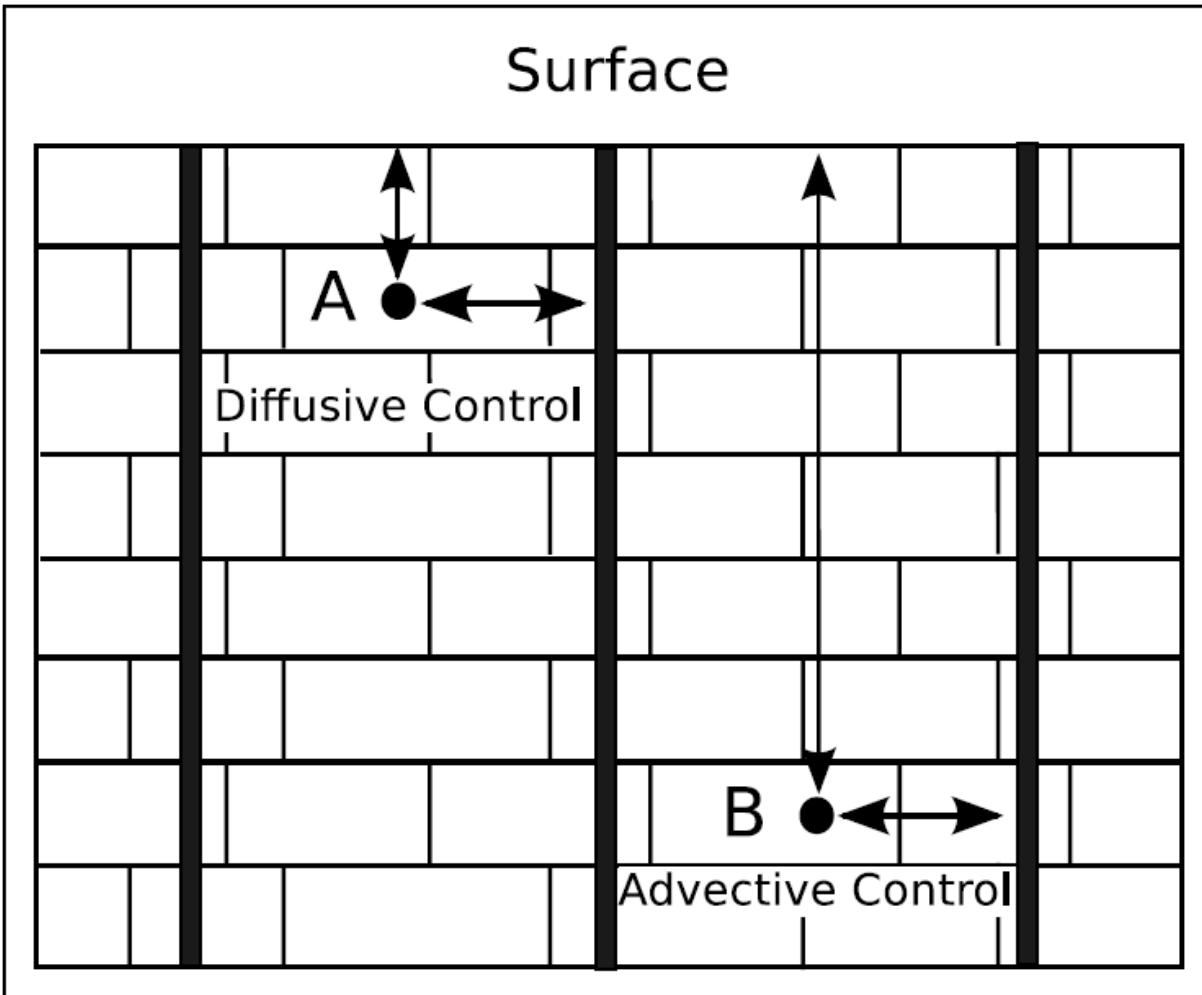
642 Figure 5. Example profiles of water and air CO₂ with depth in the vadose zone for different
 643 choices of parameters as shown in each panel. These examples assume a source of CO₂ at the
 644 bottom ($X^*=1$). The solid line depicts the air CO₂ concentration, the black dashed line depicts the
 645 water CO₂ concentration, and the grey dashed line depicts the purely diffusive profile for air
 646 CO₂. The location of each case within the parameter space is also depicted in Figure 4. The water
 647 film influences the CO₂ air concentration when $\xi > 1$ and $\xi > \Lambda_w$. When $\Lambda_w > 1$ then the CO₂ profile
 648 in the water is closely coupled to the CO₂ profile in the air.

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654 Figure 6. A profile through the vadose zone with three advectively dominated vertical fractures
 655 through the rock. At point A, dynamics will be largely controlled by diffusion, because it is as
 656 close to the surface as it is to the fractures. However, at point B advection is likely to influence
 657 CO₂ concentration, since diffusion to the advective fracture will be much faster than diffusion to
 658 the surface. Though much of the vadose zone may not be in close proximity to a fracture flow
 659 path dominated by advection, many points within the vadose zone will be closer to such a path
 660 than they are to the surface.