

1 The importance of advection for CO<sub>2</sub> dynamics in the karst

2 Critical Zone: an approach from dimensional analysis

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5 **ABSTRACT**

6 Karst landscapes provide unique challenges and opportunities to studies of processes  
7 within the Critical Zone, which spans from the top of the canopy to the base of active  
8 groundwater circulation. Dimensional analysis using the characteristic length scales and times  
9 scales of karst processes enables development of an initial framework for quantification of the  
10 rates and distribution of these processes throughout the Critical Zone. In particular, dimensional  
11 analysis provides a useful tool to identify the relative importance of different processes and to  
12 test the assumptions behind models of Critical Zone function. I briefly review prior use of  
13 dimensional analysis to understand various aspects of the karst Critical Zone and then introduce  
14 simple models for CO<sub>2</sub> transport within fractures and conduits of the vadose zone. Dimensional  
15 analysis of these models suggests that advection through karst fractures within both the gas and  
16 liquid phases will strongly impact vertical CO<sub>2</sub> profiles in the vadose zone in a variety of  
17 settings. Implications of this finding for karst Critical Zone development and future data  
18 campaigns are discussed.

19 **INTRODUCTION**

20 The Critical Zone (CZ) spans from the top of the canopy to the base of active  
21 groundwater circulation (Brantley et al., 2006). Though an understanding of CZ processes is  
22 crucial to predicting how Earth's surface will respond to climate change and anthropogenic

23 impacts, and to unraveling the connections between the physical, chemical, and biological  
24 processes that drive landscape evolution, CZ processes and the coupling between them remain  
25 relatively poorly understood. This has led to recent efforts toward a general and systematic  
26 approach to CZ research as spearheaded by the Critical Zone Exploration Network (CZEN), a  
27 network of field sites and community of researchers focused on understanding CZ processes.

28 Karst landscapes, which form in settings with highly soluble lithologies and occupy  
29 roughly 12% of Earth's surface (Ford and Williams, 2007), provide an end-member case of CZ  
30 dynamics. In contrast to other settings, karst systems evolve largely by congruent weathering.  
31 Additionally, karst systems allow extremely rapid infiltration of water and the solutes and  
32 particles transported by water. In karst settings, it is possible to have freshly recharged and  
33 chemically aggressive water circulating at depths of hundreds or even thousands of meters within  
34 minutes to hours after leaving the surface. One of the challenges of understanding CZ processes  
35 is that they span many orders of magnitude in timescales (Brantley et al., 2006). This is  
36 exacerbated in karst systems where hydraulic conductivities, and corresponding flow timescales,  
37 span over 10 orders of magnitude depending upon the length scale over which they are measured  
38 (Worthington, 2009). One advantage to karst CZ research is that, in many cases, the conduit flow  
39 network that drains much of the CZ is physically accessible to humans in the form of caves.  
40 Despite karst's unique position within the range of Earth's CZ dynamics, and relatively  
41 widespread global distribution, a karst field site is not represented within the current CZEN, and  
42 karst CZ dynamics are arguably even more poorly understood than other settings.

43 The preferential flow paths of dissolutionally enlarged fractures and conduits are of  
44 central importance to understanding the dynamics of the karst CZ as they largely control the  
45 distribution of water flow throughout the CZ and are the locations where geomorphic work is

46 focused. The distribution and apertures of such paths, and their ability to transmit fluids and  
47 weathering products, will largely determine the nature of the karst landscape. Many processes  
48 occurring within the CZ are controlled by gradients in temperature, solute concentration, carbon  
49 dioxide, and the availability of dissolved and particulate organic matter. To understand the  
50 variability of these quantities with depth, and as a function of flow path properties, the concept of  
51 a process length scale is a useful tool. If a given process that occurs over a characteristic time  
52 scale is coupled to flow along a karst fracture or conduit, then the combination of this  
53 characteristic time scale with the flow velocity leads to a characteristic length scale (Covington  
54 et al., 2012). These length scales provide a quantitative tool for dividing different dynamic  
55 regimes, and for estimating the decay of critical zone processes with depth.

56         Process length scales are a specific example of dimensional analysis, whereby  
57 characteristic length scales, time scales, and relevant dimensionless numbers are derived from  
58 the equations governing a process. Here, I summarize previous use of such methods to  
59 understand processes relevant to karst CZ dynamics and then use this framework to generate  
60 questions and hypotheses concerning karst CO<sub>2</sub> dynamics and to test assumptions of current  
61 models.

## 62 **DIMENSIONAL ANALYSIS OF PROCESSES IN THE KARST CRITICAL ZONE**

### 63 **Heat Transport**

64         Temperature is an important control on many of the chemical reactions that drive CZ  
65 processes, including dissolution and microbial production of CO<sub>2</sub>. Temperature variations at  
66 Earth's surface propagate into the soil and rock and penetrate to a depth that is often referred to  
67 as the thermal penetration depth, or skin depth (Incropera et al., 2007, pg. 299). This depth is  
68 dependent on the time scale of temperature variations and is given by a relationship of the form

69  $\lambda_{skin} \sim \sqrt{D_{th}t},$  (1)

70 where  $D_{th}$  is the thermal diffusivity of rock, and  $t$  is the timescale of temperature variations.  
71 Fluxes of air and water may enhance the depth of surface influence on soil and rock temperatures  
72 within the subsurface. Specifically, karst conduits provide pathways that exchange both air and  
73 water with the surface. Luetscher and Jeannin (2004) found that flow of both of these fluids can  
74 result in substantial heat exchange within a karst system, and that air temperature with depth in a  
75 karst aquifer normally trends with the adiabatic lapse rate, demonstrating the importance of air  
76 circulation. Wigley and Brown (1971) derived characteristic length scales associated with  
77 equilibration of atmospheric air to cave conditions. Using equations for the exchange of heat  
78 between fluid and rock, one can derive the thermal length scale associated with fluid flow down  
79 a fracture or conduit within the subsurface (Covington et al. 2012). All of these relationships  
80 allow relatively simple calculations of the depth to which thermal variations should penetrate  
81 into the karst CZ under different conditions. Roughly speaking, conductive heat changes  
82 penetrate a few meters into the subsurface, air temperature changes penetrate a few hundreds of  
83 meters, and water temperature changes can penetrate pathways multiple kilometers in length,  
84 though relatively large (meter scale) conduits are needed for air and water temperature changes  
85 to reach such distances into the subsurface.

## 86 **Dissolution**

87 The most widely discussed length scale in karst is the dissolutional penetration length  
88 (Weyl, 1958; Dreybrodt, 1990; Palmer, 1991), the distance that undersaturated water will  
89 penetrate along a flow path before approaching saturation. Within a linear kinetic regime, and at  
90 a constant flow velocity, water will approach saturation exponentially over a length scale

91  $\lambda_d = \frac{VD_H}{4\alpha_s},$  (2)

92 where  $V$  is the flow velocity,  $D_H$  is the hydraulic diameter of the flow path, and  $\alpha_s$  is the linear  
93 kinetic rate constant (for a derivation see Covington et al., 2012). However, calcite dissolution  
94 kinetics become non-linear as saturation is approached, allowing slightly undersaturated water to  
95 penetrate much deeper than it would otherwise (Dreybrodt, 1990; Palmer, 1991). Consideration  
96 of the exponential length scale in Equation 2 allows an approximation of the distance over which  
97 most of the aggressivity of the water is spent. Gabrovšek (2007; 2009) examined the vertical  
98 distribution of dissolution rates as water percolates through the vadose zone. This analysis was  
99 used to explore the relationship between measured solute fluxes and surface denudation rates  
100 within a karst basin. Dissolution length scales have also been compared against CO<sub>2</sub> degassing  
101 length scales and matrix inflow length scales to aid interpretation of observed downstream  
102 decreases of dissolution rates in a cave stream (Covington et al., 2013). In general, dissolution  
103 length scales are quite long (tens to hundreds of km) in mature karst conduits (Covington et al.,  
104 2012), so changes in dissolution rates along a conduit are more likely to be a function of CO<sub>2</sub>  
105 dynamics or mixing of waters with different chemistries than dissolution occurring along the  
106 flow path.

### 107 **Carbon Dioxide Dynamics**

108 Carbon dioxide, as the primary source of aggressivity in karst waters, provides an  
109 important control on the distribution and rate of weathering processes within the karst Critical  
110 Zone. CO<sub>2</sub> is produced within the soil zone due to root respiration and oxidation of organic  
111 material. However, data from a number of sites also demonstrate that CO<sub>2</sub> concentrations can  
112 increase with depth to values much higher than found in the soil zone, suggesting the presence of  
113 deeper sources of CO<sub>2</sub> in some settings, likely as the result of organic matter decay deeper within  
114 the vadose zone or at the water table (Atkinson, 1977; Wood and Petraitis, 1984; Baldini et al.,

115 2006; Whitaker and Smart, 2007a; Benavente et al., 2010). Models of vadose zone CO<sub>2</sub> profiles  
116 within karst are typically based on diffusive transport equations (e.g. Wood, 1985; Breecker et  
117 al., 2012), that assume that molecular diffusion is the primary mechanism for transport of CO<sub>2</sub>  
118 produced in the subsurface. Diffusive flux is described by Fick's 1st Law, which at steady state  
119 gives

$$120 \quad F_{CO_2} = \frac{D_a}{L} [(CO_2)_L - (CO_2)_0], \quad (3)$$

121 where  $F_{CO_2}$  is the flux of CO<sub>2</sub>,  $D_a$  is the diffusion coefficient of CO<sub>2</sub> in air,  $L$  is the depth of the  
122 CO<sub>2</sub> source, and  $(CO_2)_L$  and  $(CO_2)_0$  are the concentrations of CO<sub>2</sub> at depth  $L$  and the surface,  
123 respectively. An attractive feature of diffusive models is that they can explain relatively high  
124 CO<sub>2</sub> concentrations at depth. At equilibrium, the flux of CO<sub>2</sub> out of the vadose zone must be  
125 equal to the production rate within it. Therefore, even a modest production of CO<sub>2</sub> at depth will  
126 result in an increase in CO<sub>2</sub> concentration at depth until a sufficient gradient is obtained to  
127 sustain the required flux (Wood and Petraitis, 1984). However, the dissolutionally enlarged  
128 pathways within a karst setting can allow for advective transport of species in both the liquid and  
129 gaseous phases. A number of authors have noted that cave ventilation can be an important  
130 control on CO<sub>2</sub> within a cave, and consequently on the timing and magnitude of calcite  
131 precipitation on speleothems (Spötl et al., 2005; Wong et al., 2011; Frisia et al., 2011; Breecker  
132 et al., 2012; Fairchild and Baker, 2012). However, less consideration has been given to whether  
133 advective processes influence CO<sub>2</sub> dynamics within smaller aperture flow paths, such as  
134 solutionally enlarged fractures. Below, I use dimensional analysis of simple mathematical  
135 models of these processes to examine the potential influence of advection.

### 136 *Advection of gaseous CO<sub>2</sub>*

137 Since transport of CO<sub>2</sub> in the gas phase through the vadose zone is typically assumed to  
138 be diffusive, yet karst vadose zones contain relatively open pathways that may allow advection  
139 to occur, I will first examine the conditions under which one might expect for advection of gases  
140 along fractures to be important. A number of mechanisms can drive cave airflow, although  
141 arguably the most important type of cave airflow is chimney-effect airflow (Wigley and Brown,  
142 1976), whereby temperature-related contrasts between surface and subsurface air density drive  
143 buoyant flow between two or more entrances at different elevations. While this airflow  
144 mechanism is well-documented within human-enterable caves, an interesting question is whether  
145 it might drive significant flow within the fracture systems as well. The pressure difference,  $\Delta P$ ,  
146 that drives chimney effect airflow can be approximated as

$$147 \quad \Delta P = \rho_{in} g h \frac{\Delta T}{T_{ext}}, \quad (4)$$

148 where  $\rho_{in}$  is the density of air inside the cave or fracture,  $g$  is Earth's gravitational acceleration,  $h$   
149 is the elevation difference between the two openings to the surface,  $\Delta T$  is the temperature  
150 difference between the subsurface and surface air, and  $T_{ext}$  is the surface air temperature in  
151 Kelvin (e.g. Badino, 2010). Considering laminar flow within a fracture, the flow velocity is  
152 given by the Hagen-Poiseuille equation,

$$153 \quad V = \frac{\Delta P a^2}{12 \mu_a L}, \quad (5)$$

154 Where  $V$  is the flow velocity,  $a$  is the fracture aperture,  $\mu_a$  is the dynamic viscosity of air, and  $L$   
155 is the length of the flow path. For turbulent flow conditions, the flow velocity is given by the  
156 Darcy-Weisbach equation

$$157 \quad V = \sqrt{\frac{4a\Delta P}{fL}}, \quad (6)$$

158 where  $f$  is the Darcy-Weisbach friction factor. The Peclet Number (Pe), which is the ratio of  
159 diffusive and advective time scales, is used to quantify the relative importance of diffusion and  
160 advection,

$$161 \quad \text{Pe} = \frac{VL}{D}, \quad (7)$$

162 where  $D$  is the diffusion coefficient for the species of interest and  $L$  is the length scale of interest,  
163 which is chosen here to be equal to the flow path length (Incropera et al., 2007). When  $\text{Pe} \gtrsim 1$   
164 advective transport is important, and when  $\text{Pe} \gg 1$  then transport is dominated by advection  
165 rather than diffusion (Fetter, 1993). Note that, unless flow velocity is zero, there will always  
166 exist a length scale,  $L$ , above which advection is dominant. For laminar flow, Equations 4, 5, and  
167 7 can be combined to give

$$168 \quad \text{Pe}_{a,\text{lam}} = \frac{\rho_{in} g h a^2 \Delta T}{12 \mu_a D_a T_{ext}}. \quad (8)$$

169 If one uses the flow path length,  $L$ , as the characteristic scale for which Pe is calculated, then for  
170 the laminar flow case Pe is independent of flow path length. For turbulent flow, Equations 4, 6,  
171 and 7 are combined giving

$$172 \quad \text{Pe}_{a,\text{turb}} = \frac{L^{1/2}}{D_a} \sqrt{\frac{4 \rho_{in} g h a \Delta T}{f T_{ext}}}. \quad (9)$$

173 For turbulent flow, Pe does depend on flow path length, and therefore calculation of Pe requires  
174 more information about the geometry of the flow path than in the laminar case.

175 To quantify the potential importance of advective processes within fractures in the karst  
176 vadose zone, I calculate Pe for a range of fracture apertures,  $a$ , and for entrance elevation  
177 differences of 0.3, 3, 30, and 300 m (Figure 1). A temperature ratio (in Kelvin) of  $\Delta T/T_{ext} =$   
178 3.5% is assumed, which corresponds roughly to a difference between surface and subsurface  
179 temperatures of 10°C with a subsurface temperature of 10°C. This choice is well within the



180 expected range for natural systems. Air properties are also taken at 10°C with  $\rho_{in} = 1.25 \text{ kg m}^{-3}$ ,  
181  $\mu_a = 1.8 \times 10^{-5} \text{ (N}\cdot\text{s) m}^{-2}$ , and  $D_a = 1.55 \times 10^{-5} \text{ m}^2\text{s}^{-1}$ . For turbulent flow cases, a value of  
182  $f = 0.1$  is used, though the conclusions do not depend strongly on this choice since all of the  
183 turbulent flow cases are well within the advective regime. For turbulent flow, the flow path  
184 length influences Pe, and therefore two example cases are chosen. In the first case (Figure  
185 1A,B),  $L$  is set equal to the difference between the two entrance elevations,  $h$ . This is a limiting  
186 case, as  $L$  cannot be any shorter. However, it provides a reasonable approximation to a vertical  
187 fracture that intersects a horizontal cave (Figure 1A). For the second case  $L=100 h$ , which  
188 represents a situation where the flow path length is much greater than the elevation difference  
189 between the two fracture outlets (Figure 1C,D).

190         Figure 1 suggests that advection should be important in a wide variety of cases. For  
191 example, even with a small elevation difference of 0.3 m, advection is an important process  
192 within fractures with apertures on the order of 0.1 mm and larger. Differences between the short  
193 (Figure 1A,B) and long flowpath (Figure 1C,D) cases are relatively minor because of: 1) the  
194 independence of the laminar Pe on flow path length, and 2) the weak dependence of Pe on flow  
195 path length in the turbulent regime. For the long flow path case, the transition to turbulence  
196 occurs at slightly larger apertures and Pe, and the turbulent flow Pe values are also a factor of 10  
197 larger. This results because of the positive effect of length on Pe that is only partially offset by  
198 additional flow resistance.

199         The above analysis considers fractures that are open to the surface. While exposed  
200 fractures are likely to occur in many karst settings, often the rock will be covered by a layer of  
201 soil or regolith. Such a cover can substantially influence the extent of air exchange through the

202 fracture network below. If the cover behaves as a porous medium, then the specific discharge of  
 203 air through the cover is given by Darcy's Law,

$$204 \quad q = \frac{\kappa \Delta P_p}{\mu_a L_p}, \quad (10)$$

205 where  $q$  is the specific discharge,  $\kappa$  is the permeability,  $\Delta P_p$  is the pressure drop across the  
 206 porous cover, and  $L_p$  is the thickness of the cover. If the porous cover and fracture are connected  
 207 in series, then one can assume that the flow velocity in the fracture is equal to the specific  
 208 discharge in the cover and that the total chimney effect pressure drop (Equation 4) is equal to the  
 209 sum of the pressure drops across the cover and fracture. Using these assumptions, assuming  
 210 laminar flow, and again setting the characteristic length scale to the fracture flow path length,  
 211 leads to a new relation for Pe inside the fracture in the case of cover,

$$212 \quad \text{Pe}_{a,\text{cover}} = \frac{\rho_{in} g h}{\mu_a D_a} \frac{\Delta T}{T_{ext}} \left( \frac{1}{12/a^2 + F/\kappa} \right), \quad (11)$$

213 where  $F$  is the ratio of cover thickness to fracture length. There are two limiting cases for  
 214  $\text{Pe}_{a,\text{cover}}$ . When  $\frac{12}{a^2} \gg \frac{F}{\kappa}$ , then the bulk of the pressure loss is in the fracture, the cover has little  
 215 effect, and Equation 8 applies. Alternatively, when  $\frac{12}{a^2} \ll \frac{F}{\kappa}$ , then the bulk of the pressure loss is in  
 216 the cover, and the fracture properties no longer influence Pe. Comparing the magnitude of these  
 217 two terms allows estimation of the strength of the influence of the cover. In the limit where the  
 218 pressure loss is entirely in the cover, Pe approaches a maximum possible value of

$$219 \quad \text{Pe}_{a,\text{max}} = \frac{\kappa \rho_{in} g h}{\mu_a D_a F} \frac{\Delta T}{T_{ext}}. \quad (12)$$

220 Two example cases are shown in Figure 2, where  $F=0.05$  and  $\kappa$  is set to  $10^{-9} \text{ m}^2$  in the first case  
 221 (Figure 2A) and  $10^{-12} \text{ m}^2$  in the second (Figure 2B). One can see that there are realistic cases  
 222 where the cover would either allow advection to dominate or would shut it down. This model  
 223 likely produces an underestimate of the importance of advection, since it assumes that the air

224 exiting or entering the fracture will pass through a column of cover equal in cross sectional area  
225 to the fracture. More likely, air would spread through a cone within the cover and perhaps  
226 channel along preferential flow paths due to heterogeneity within the cover.

### 227 *Advection of water and its influence on CO2*

228 In the previous section I examined whether advection within the gaseous phase may play  
229 an important role in the karst vadose zone. However, even in the absence of gas advection along  
230 fractures and conduits, water is channeled along these flow paths, particularly during wet periods  
231 or following recharge events. The water and air do not evolve in isolation but rather CO<sub>2</sub> will  
232 exchange between the two. Here I will examine whether advective flows of water can influence  
233 an otherwise diffusive CO<sub>2</sub> profile within a fracture. To examine this question, I use coupled  
234 advection-dispersion equations for air and water within a fracture,

$$235 \quad \frac{\partial(\text{CO}_2)_w}{\partial t} = D_w \frac{\partial^2(\text{CO}_2)_w}{\partial x^2} - V_w \frac{\partial(\text{CO}_2)_w}{\partial x} - \frac{D_w}{\delta_w^2} [(\text{CO}_2)_w - \alpha(\text{CO}_2)_a] + R, \quad (13)$$

$$236 \quad \frac{\partial(\text{CO}_2)_a}{\partial t} = D_a \frac{\partial^2(\text{CO}_2)_a}{\partial x^2} - V_a \frac{\partial(\text{CO}_2)_a}{\partial x} + \frac{D_w}{\delta_w \delta_a} [(\text{CO}_2)_w - \alpha(\text{CO}_2)_a], \quad (14)$$

237 where  $t$  is time,  $x$  is position along the fracture,  $(\text{CO}_2)_a$  and  $(\text{CO}_2)_w$  are CO<sub>2</sub> concentrations in the  
238 air and water, respectively,  $D_w$  is the diffusion coefficient for CO<sub>2</sub> in water,  $V_w$  and  $V_a$  are the  
239 flow velocities of air and water, respectively,  $\delta_w$  is the thickness of the water film on the fracture  
240 wall,  $\delta_a$  is the aperture of the air-filled portion of the fracture,  $\alpha$  is the Ostwald solubility  
241 coefficient for CO<sub>2</sub> (Wanninkhof et al., 2009), and  $R$  is a term representing CO<sub>2</sub> production  
242 within the water or at the water-rock boundary. The geometry of this model is depicted in Figure  
243 3. The exchange terms (the third terms on the right hand side of each equation) assume that CO<sub>2</sub>  
244 exchange is controlled by the rate of diffusion of CO<sub>2</sub> through the water film. This simple model  
245 is arguably a gross oversimplification of dynamics in karst fractures, but dimensional analysis of  
246 the model will allow an initial examination of assumptions and elucidation of relevant process

247 controls. One process not represented in Equations 13 and 14 is the interaction between CO<sub>2</sub>  
 248 concentrations and dissolution or precipitation of calcite. These processes are discussed briefly  
 249 below, and can easily be introduced with an additional term, but are excluded here for the sake of  
 250 brevity.

251 To examine whether flowing water might influence CO<sub>2</sub> profiles within the vadose zone,  
 252 I consider a case where air velocity is zero, such that the CO<sub>2</sub> profile, if unperturbed by the flow  
 253 of water, would be controlled by diffusion. Setting the time derivative terms to zero allows us to  
 254 examine steady state CO<sub>2</sub> profiles, and for simplicity I also consider a case with no CO<sub>2</sub>  
 255 production ( $R=0$ ). To non-dimensionalize the model, I define a dimensionless position variable  
 256  $x^* = x/L$ , and dimensionless concentration,  $(CO_2^*)_i = (CO_2)_i / (CO_2)_0$ , where  $L$  is a characteristic  
 257 length scale, and  $(CO_2)_0$  is a reference concentration of CO<sub>2</sub>. In the example solutions below,  
 258  $(CO_2)_0$  is set to the standard atmospheric concentration of CO<sub>2</sub>, though any reference value could  
 259 be used. Using these definitions and assumptions, Equations 13 and 14 can be re-written as

$$260 \quad \frac{1}{Pe_w} \frac{\partial^2 (CO_2^*)_w}{\partial x^{*2}} = \frac{\partial (CO_2^*)_w}{\partial x^*} + \Lambda_w [(CO_2^*)_w - \alpha (CO_2^*)_a], \quad (15)$$

$$261 \quad \frac{\partial^2 (CO_2^*)_a}{\partial x^{*2}} = -\xi [(CO_2^*)_w - \alpha (CO_2^*)_a], \quad (16)$$

262 where  $Pe_w = V_w L / D_w$ ,  $\xi = D_w L^2 / D_a \delta_w \delta_a$ , and  $\Lambda_w = D_w L / \delta_w^2 V_w$  are all dimensionless numbers  
 263 that determine the nature of the solution.  $Pe_w$  is the Peclet Number for transport within the water  
 264 and determines the relative importance of advection and diffusion along the flowing film.  $\xi =$   
 265  $(D_w L / \delta_w) / (D_a \delta_a / L)$  represents the ratio between the water-air CO<sub>2</sub> exchange rate and the  
 266 diffusive mass transport rate within the air along the length of the fracture. When  $\xi$  is large, then  
 267 exchange of CO<sub>2</sub> between the water and air will be important in comparison to diffusion in the  
 268 air along the fracture. When  $\xi$  is small then exchange is insignificant in comparison to diffusive  
 269 transport. The final dimensionless number,  $\Lambda_w = L / \lambda_w$ , is a ratio of the fracture length and the

270 length scale over which the water CO<sub>2</sub> concentration equilibrates with the air,  $\lambda_w = V_w \delta_w^2 / D_w$ . If  
271  $\Lambda_w$  is small, then the water CO<sub>2</sub> concentration will be decoupled from the air. If  $\Lambda_w$  is large, then  
272 the CO<sub>2</sub> concentration in the water will remain nearly equilibrated to that of the air over length  
273 scales greater than  $L$ . These dimensionless numbers are summarized in Table 1.

274 (Table 1 near here)

275 In order for flow of water through the vadose zone to influence CO<sub>2</sub> profiles within the  
276 air, two physical conditions must hold: 1) the rate of transfer of CO<sub>2</sub> between air and water must  
277 be comparable to or greater than the rate of diffusive transport within the air, otherwise any  
278 perturbations to the profile via exchange will be efficiently smoothed by diffusion in the air, and  
279 2) the advective mass transport rate of CO<sub>2</sub> in the water film must be comparable to or greater  
280 than the diffusive mass transport rate in the air, otherwise the water will not have the transport  
281 capacity to influence CO<sub>2</sub> concentrations in the air. The first condition can be quantified as  $\xi \gtrsim$   
282 1. The second condition is met when the ratio  $\xi / \Lambda_w = \delta_w V_w / (\frac{D_a \delta_a}{L}) \gtrsim 1$ , as this ratio expresses  
283 the ratio of the mass transport rate via advection in the water to the mass transport rate via  
284 diffusion in the air. For  $\xi / \Lambda_w < 1$  the diffusive transport in the air outpaces advective transport  
285 in the water, and the film has no significant affect. This occurs in the limit of low film velocity  
286 and thin films.

287 To determine whether downward flow of water is likely to influence air CO<sub>2</sub> profiles in  
288 nature, we can examine typical values of the dimensionless numbers  $\xi$  and  $\Lambda_w$  under conditions  
289 that occur in natural systems. Concerning  $\xi$ , the ratio of the diffusion coefficients for CO<sub>2</sub> in air  
290 and water is roughly constant,  $D_w / D_a \approx 10^{-4}$ . Therefore we need only consider the other  
291 variables,  $L$ , which can be taken to be the thickness of the vadose zone or the length of the  
292 fracture flow path, and  $\delta_a$  and  $\delta_w$ , which are the aperture of the air-filled fracture and the

293 thickness of the water film, respectively. For a thin film of water flowing on a vertical wall, the  
294 thickness of the film can be approximated as (Bird et al., 2002)

$$295 \quad \delta_w = \sqrt{\frac{3V_w\mu_w}{g\rho_w}}. \quad (17)$$

296 Using this relationship and the ratio of diffusion coefficients above, the equations for  $\zeta$  and  $\Lambda_w$   
297 can be rewritten in terms of  $V_w$  as

$$298 \quad \zeta = \frac{10^{-4}L^2}{\delta_a} \sqrt{\frac{g\rho_w}{3V_w\mu_w}} \quad (18)$$

$$299 \quad \Lambda_w = \frac{LD_w g\rho_w}{3V_w^2\mu_w}. \quad (19)$$

300 To examine the likely ranges of these parameters in nature, I calculate  $\zeta$  and  $\Lambda_w$  for  
301 fractures/conduits with air-filled apertures of 0.001 m and 0.1 m and for vadose zone thicknesses  
302 of  $L = 0.3, 3, 30,$  and  $300$  m. The dimensionless parameters are calculated as a function of  $V_w$   
303 over a range from  $10^{-6} \text{ m s}^{-1}$  to  $10 \text{ m s}^{-1}$  (Figure 4).  $\zeta$  is small for only a few cases where the  
304 vadose zone is very thin and the flow very fast. Therefore the ratio  $\zeta/\Lambda_w$ , which expresses the  
305 relative rates of mass transport in the water and air, is the primary determining factor for whether  
306 the water influences the air  $\text{CO}_2$  profile. The water advective influence is stronger for thicker  
307 vadose zones and when the air-filled aperture is smaller (Figure 4A). However, in most settings  
308 air  $\text{CO}_2$  profiles are influenced by water advection for flow velocities above about  $1 \text{ mm s}^{-1}$ ,  
309 which is well within the range of expected velocities in natural fractures.

310 To illustrate the influence of these dimensionless numbers on the shape of equilibrium  
311  $\text{CO}_2$  profiles, and the likely importance of advection, I numerically solve Equations 15 and 16 in  
312 Python using the free and open source finite volume package FiPy (Guyer et al., 2009). The  
313 equations are solved at steady-state for the boundary conditions

$$314 \quad \frac{1}{\alpha} (\text{CO}_2^*)_w|_{x=0} = (\text{CO}_2^*)_a|_{x=0} = 1, \quad (20)$$

315  $(\text{CO}_2^*)_a|_{x=L} = 10$  (21)

316  $\frac{d(\text{CO}_2^*)_w}{dx}|_{x=L} = 0.$  (22)

317 The first two boundary conditions (20) set the air and water  $\text{CO}_2$  at the top of the fracture to  
318 values that are equilibrated with the atmosphere. The third condition (21) sets the  $\text{CO}_2$   
319 concentration at depth  $L$  equal to 10 times atmospheric, and the final condition (22) is an outlet  
320 boundary condition at the downstream end of the water film. These boundary conditions  
321 represent the case of a  $\text{CO}_2$  source at depth  $L$ , which is only one of many possibilities. For  
322 example, profiles would look quite different for a source near the surface. However, solutions for  
323 this example set of boundary conditions will illustrate how the dimensionless parameters, and the  
324 relative importance of physical processes that they represent, influence the form of the solution.  
325 These conclusions also hold for other conditions, such as a near-surface  $\text{CO}_2$  source, although the  
326 shapes of the profiles will, in general, be different.

327 Solutions are shown for four example cases that span a variety of possible combinations  
328 of the dimensionless parameters. Individual profiles are shown in Figure 5 and their location on  
329 Figure 4 is also marked with identifying numbers. When  $\zeta$  is small, the water film has no  
330 significant influence on the air  $\text{CO}_2$ , and the profile closely matches that of the purely diffusive  
331 case (Figure 5A). If  $\zeta > 1$  and  $\Lambda_w < 1$  then the  $\text{CO}_2$  air profile is strongly influenced by the water,  
332 but the water film  $\text{CO}_2$  concentration is not strongly affected by the air  $\text{CO}_2$  as the water film is  
333 moving too quickly to be significantly influenced (Figure 5B). When both  $\zeta$  and  $\Lambda_w$  are  
334 significantly larger than one and  $\xi/\Lambda_w \gtrsim 1$ , then the air  $\text{CO}_2$  profile is significantly influenced  
335 by the water film, but the water film stays near equilibrium with the air at all depths (Figure 5C).  
336 Finally, when  $\xi \gg 1$  and  $\xi/\Lambda_w \lesssim 1$  then the  $\text{CO}_2$  profile is nearly diffusive and the water profile

337 is close to equilibrium with the air, because the water film has insufficient transport capacity in  
338 comparison to diffusion rates in the air (Figure 5D).

## 339 **DISCUSSION AND CONCLUSIONS**

340 While the models presented here are simple, and natural karst systems undoubtedly  
341 display a broader and richer range of dynamics, these simple models provide an initial means of  
342 thinking about the relative importance of different processes and the likely interactions between  
343 them. In the past, vadose zone gas dynamics has typically been treated as diffusive, where  
344 advective transport, within both the gas and liquid phases, is neglected (Penman, 1940a; Penman,  
345 1940b; Marshall, 1959; Wood and Petraitis, 1984). Buoyancy driven flows have received recent  
346 consideration within fracture systems (Weisbrod and Dragila, 2006; Weisbrod et al., 2009;  
347 Nachshon et al., 2008). However, in karst systems dissolutional enlargement of fractures may  
348 further increase the importance of advective processes.

349 In the models presented above, I first examine the possibility of advective air flows that  
350 are driven by chimney effect pressure gradients. The influence of cave air flow patterns on CO<sub>2</sub>  
351 concentrations has been well-documented (Troester and White, 1984; Ek and Gewalt, 1985;  
352 Spötl et al., 2005; Baldini et al., 2008; Milanolo and Gabrovšek, 2009; Wong et al., 2011;  
353 Breecker et al., 2012; Gulley et al., 2014), and chimney effect airflow is arguably the most  
354 common mechanism of cave airflow (Wigley and Brown, 1976). Previous authors have  
355 demonstrated, using similar dimensional analysis, that advection should be a dominant process  
356 for CO<sub>2</sub> transport in human-enterable caves (Frisia et al., 2011; Fairchild and Baker, 2012).  
357 However, the dimensional analysis above suggests that chimney effect airflow should also be  
358 important in many fractures or small solutionally enlarged pathways. For example, a fracture  
359 with a 1 mm aperture that has two entrances with an elevation difference of only 0.3 m has a



360 Peclet Number of more than 100 in the case of a surface temperature 10 degrees different from  
361 the subsurface temperature. This indicates that transport should be predominantly advective in  
362 that case. Advection becomes even stronger with larger elevation differences. The suggested  
363 importance of advection through smaller diameter flow paths is consistent with field data. For  
364 example, Obir Cave, with no known upper entrance, displays strong chimney effect airflow  
365 (Spötl et al., 2005).

366         Since fractures that are open to the surface should experience significant airflow in a  
367 wide variety of conditions (Figure 1), the presence of significant chimney effect air flow is likely  
368 to be primarily a function of cover. If there is a thick, relatively impermeable, cover then this will  
369 shut down chimney effect airflow within the fracture network below it. The extent to which a  
370 cover impedes airflow can be approximated by calculating the theoretical maximum Peclet  
371 Number (Equation 12) appropriate for the cover thickness and permeability, though this  
372 approach is likely to somewhat overestimate the capability of the cover to halt air circulation.  
373 The fracture network pattern, and particularly the extent of horizontal connectivity, is also likely  
374 to be an important control. A fracture network with more extensive horizontal connectivity is  
375 more likely to host connected pathways that intersect the surface at substantially different  
376 elevations.

377         Even if advection dominates CO<sub>2</sub> transport within fractures and caves, one might ask  
378 whether this influence exists only in isolated zones in the vicinity of these preferential flow  
379 paths. It may be the case that some portions of the vadose zone are isolated from the effects of  
380 advection. However, diffusion occurs both vertically and laterally. Therefore the influence of  
381 advection at a particular point within the vadose zone is likely to depend on the ratio of the  
382 distance between that point and the surface and that point and the nearest advection-dominated

383 path (Figure 6). If, at a given location, the nearest advection-dominated path is closer than the  
384 surface, then advection is likely to have an important influence at that location. Of course, any  
385 anisotropy between vertical and lateral diffusion rates must also be accounted for.

386 Chimney effect airflow implies that there are regions of both inflow and outflow of air  
387 that vary with season and surface temperature. Therefore, CO<sub>2</sub> dynamics can be expected to vary  
388 diurnally and seasonally. Lower and upper entrances will also experience substantially different  
389 CO<sub>2</sub> dynamics and gradients with depth, since each undergoes the opposite direction of flow at  
390 any given time. Such effects may have important implications for the temporal and spatial  
391 distribution of geomorphic work within the karst vadose zone. In comparison to chimney effect  
392 airflow, karst systems dominated by barometric air exchange would be expected to display quite  
393 different distributions of CO<sub>2</sub> in space and time.

394 The model presented for the coupling between water and air CO<sub>2</sub> concentrations suggests  
395 that flowing water, even in the absence of air flow, is likely to alter CO<sub>2</sub> profiles away from  
396 purely diffusive profiles to curved profiles that display lower gradients near the surface and  
397 higher gradients at depth. The exact shape will depend on the boundary conditions at the water  
398 table and the distribution of CO<sub>2</sub> production with depth. In particular, the example cases shown  
399 in Figure 5 assume CO<sub>2</sub> production is located at depth. Different distributions of CO<sub>2</sub> production  
400 can result in quite different vertical CO<sub>2</sub> profiles than the ones depicted here. The strength of the  
401 advective influence of water depends on several parameters, but, for most of the parameter space  
402 sampled, water flow velocities of 1 mm s<sup>-1</sup> and greater typically result in a CO<sub>2</sub> profile that is  
403 modified by water advection. Within natural systems, such effects are likely to be event driven,  
404 and the profile may evolve seasonally and with individual recharge events. The ability for a  
405 diffusive profile to recover between recharge events will be largely dependent on the ratio of the

406 diffusion time scale for the vadose zone thickness to the time period between recharge events  
407 that modify the profile.

408         There are a variety of processes that are not considered explicitly here, but for which a  
409 similar approach may enable fruitful progress. One might ask whether there are typical length  
410 scales that govern CO<sub>2</sub> production within the subsurface. For example, roots may penetrate to  
411 typical depths that vary by plant species and other parameters. Filtration and processing of  
412 particulate and dissolved organic matter may also have associated length scales that depend upon  
413 rates of microbial decay, soil porosity and permeability, fracture aperture, recharge rate, and  
414 temperature. Understanding these controls will substantially improve our ability to predict the  
415 distribution of CO<sub>2</sub> within karst Critical Zones in different settings. Any length scales associated  
416 with CO<sub>2</sub> production could be compared against length scales of water-air CO<sub>2</sub> exchange to  
417 quantify the influence that production would have on the overall profile.

418         Similarly, comparing dissolutional length scales with CO<sub>2</sub> exchange and production length  
419 scales would improve our understanding of the distribution of geomorphic work throughout the  
420 karst Critical Zone. Under what conditions is aggressivity largely spent near the zone where it is  
421 produced versus penetrating much greater depths before the CO<sub>2</sub> has reacted? In cases where  
422 dissolutional length scales are much shorter than CO<sub>2</sub> exchange length scales then dissolution  
423 and/or precipitation is also likely itself to be an important driver of CO<sub>2</sub> dynamics. Similar  
424 coupling between wind-driven ventilation and dissolution/precipitation has been observed within  
425 a semi-arid karst setting (Roland et al., 2013).

426         The analysis presented here, motivated from a physics-based approach, is entirely  
427 theoretical. One might ask what implications it has for field studies of CO<sub>2</sub> dynamics in karst. I  
428 would argue that there are several important implications. First, while prior models have

429 typically assumed CO<sub>2</sub> transport in the vadose zone is diffusive, relatively simple models of  
430 buoyancy driven flow and transport within the karst vadose zone suggest that typical conditions  
431 can produce advective effects driven by flow of air or water. Therefore, interpretation of vadose  
432 zone CO<sub>2</sub> dynamics in karst should not assume from the start that diffusion is dominant and that  
433 flow of water does not perturb CO<sub>2</sub> concentrations. The model presented here also provides a  
434 mathematical starting point for quantifying advective effects and understanding the influence  
435 they may have on CO<sub>2</sub> distributions in time and space. CO<sub>2</sub> profiles under the influence of  
436 advection are more likely to contain sharp features that correlate with different air or water  
437 circulation zones and to change dynamically with changes in surface temperature and recharge  
438 rate. Such effects have been observed within a deep karst vadose zone in Spain within boreholes  
439 where air flow was also observed (Benavente et al., 2010).

440       Spatial and temporal variability in CO<sub>2</sub> may also play an important an important role in  
441 speleogenesis (Gabrovšek and Dreybrodt, 2000; Gabrovšek et al., 2000; Whitaker and Smart,  
442 2007b; Gulley et al., 2013; Gulley et al., 2014), and advective processes are likely to be an  
443 important control on the spatial and temporal distribution of dissolution within mature cave  
444 systems. The distribution of CO<sub>2</sub> within the vadose zone, the rate of exchange of CO<sub>2</sub> between  
445 air and water, and the interactions between water and rock via dissolution or precipitation all  
446 influence carbon isotope fractionation (Fairchild and Baker, 2012, p. 176-180). Therefore, the  
447 conclusions here concerning the importance of advection have important implications for studies  
448 of paleoclimate that employ speleothems.

449       In summary, the models presented here suggest that CO<sub>2</sub> dynamics within the karst Critical  
450 Zone may be controlled by a number of processes that have traditionally been neglected.  
451 Currently, there is little data available to constrain the range of dynamics that occur in natural

452 systems. Data campaigns that sample densely in both space and time may provide substantial  
453 insight into the processes that drive landform and subsurface evolution in karst terrains.

#### 454 **Acknowledgments**

455 I appreciate stimulating discussions with Matija Perne and Joe Myre during the development of  
456 this model. Insightful comments by reviewers Steffen Birk, Ian Fairchild, and Pierre-Yves  
457 Jeannin also helped to substantially clarify and sharpen this work.

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585 **TABLES**

586 Table 1. Dimensionless numbers that determine the dynamics of the coupled water-air CO<sub>2</sub>

587 transport model.

<b>Dimensionless Number</b>	<b>Physical interpretation</b>
$Pe_w = V_w L / D_w$	Relative importance of advective and dispersive mass transport within the water film. Advection dominates for large $Pe_w$ .
$\xi = D_w L^2 / D_a \delta_w \delta_a$	Relative importance of mass transfer via water-air exchange and diffusion within the air. Exchange is important for large $\xi$ .
$\Lambda_w = D_w L / \delta_w^2 V_w$	Ratio of fracture length to the length scale associated with CO <sub>2</sub> exchange. For large $\Lambda_w$ water and air CO <sub>2</sub> are closely coupled.
$\frac{\xi}{\Lambda_w} = L \delta_w V_w / D_a \delta_a$	Relative importance of advective mass transfer in the water and diffusive mass transfer in the air. When this ratio is large mass transfer in the water is important.

588

589 **Table 2.** Description of notation

<b>Symbol</b>	<b>Definition</b>
$a$	fracture aperture
$\alpha$	Ostwald solubility coefficient
$\alpha_s$	calcite dissolution kinetic rate constant
$(CO_2), (CO_2)_a, (CO_2)_w$	concentration of CO <sub>2</sub> , in air, in water
$D_a$	diffusion coefficient for CO <sub>2</sub> in air
$D_w$	diffusion coefficient for CO <sub>2</sub> in water
$D_H$	hydraulic diameter
$D_{th}$	thermal diffusivity of rock
$\delta_a$	aperture of air-filled portion of fracture
$\delta_w$	thickness of water film on fracture wall
$f$	Darcy-Weisbach friction factor
$F$	ratio of cover thickness to fracture length ( $L_p/L$ )
$F_{CO_2}$	flux rate of CO <sub>2</sub>
$g$	Earth's gravitational acceleration
$h$	difference in elevation between two entrances
$\kappa$	permeability
$L$	fracture length or characteristic length scale
$L_p$	thickness of porous cover
$\Lambda_w$	$D_w L / \delta_w^2 V_w$ , dimensionless number (see Table 1)
$\lambda_d$	dissolutional penetration length
$\lambda_w$	CO <sub>2</sub> exchange length scale

$\lambda_{skin}$	thermal skin depth
$\mu_a$	dynamic viscosity of air
$\Delta P$	pressure difference
$\Delta P_p$	pressure difference across the porous cover
Pe	Peclet Number
$\rho_{in}$	density of air inside the cave or fracture
$q$	specific discharge (of air)
$R$	CO <sub>2</sub> production rate
$\Delta T$	difference in temperature inside and outside cave or fracture
$T_{ext}$	temperature outside of cave or fracture
$t$	time
$V (V_a \text{ or } V_w)$	velocity (of air or water)
$x$	position along the fracture
$\xi$	$D_w L^2 / D_a \delta_w \delta_a$ , dimensionless number (see Table 1)

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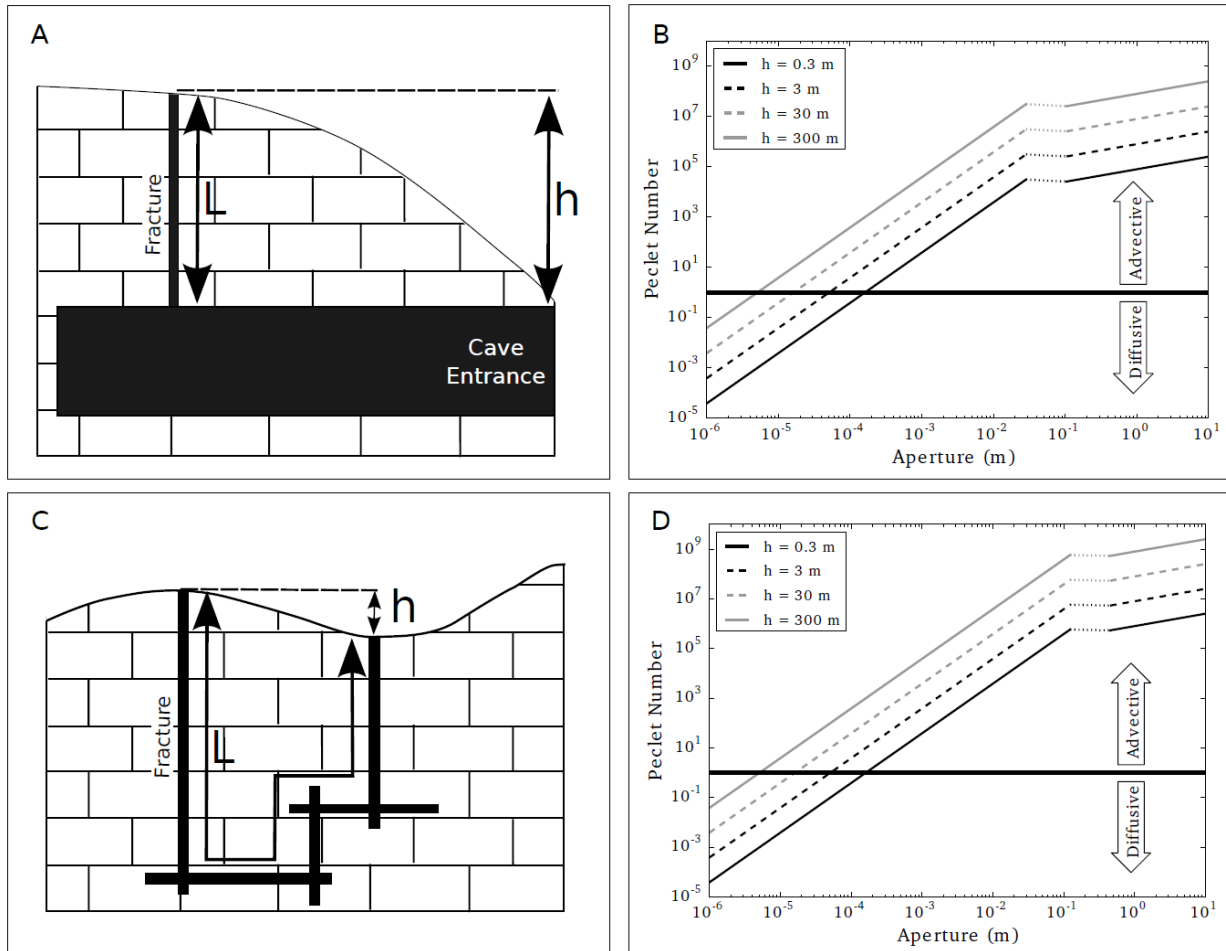
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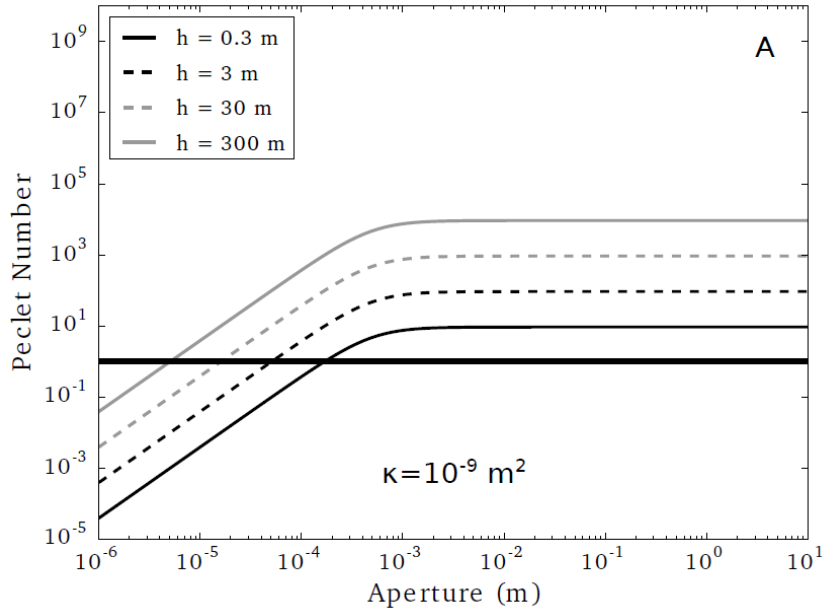
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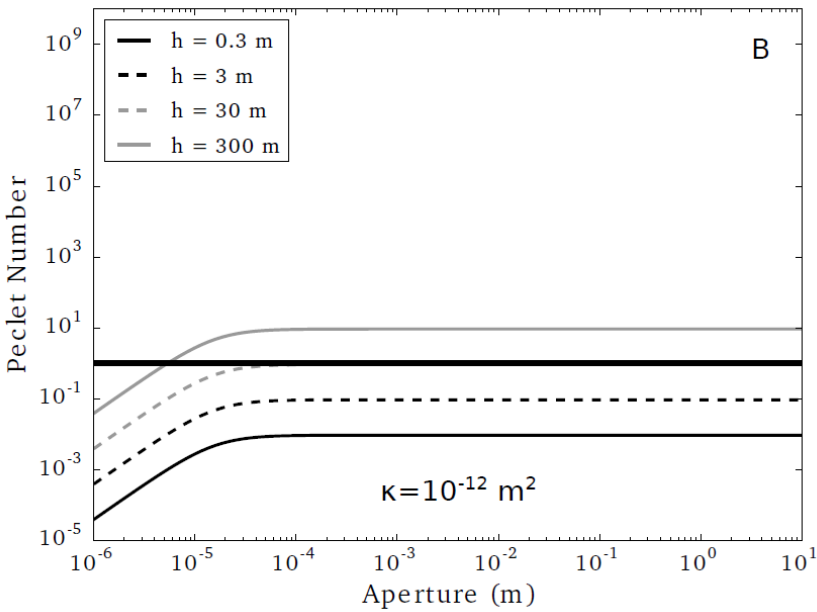
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608 Figure 1. The values of Peclet Number ( $Pe$ ) for chimney effect airflow in fractures for a range of  
 609 fracture apertures and differences between entrance elevations ( $h$ ). Transport is dominated by  
 610 advection for  $Pe \gg 1$ . Panel A shows an example geometry where the fracture length,  $L$ , is equal  
 611 to the elevation difference between entrances and  $Pe$  is calculated for this case in Panel B. Panel  
 612 C shows a case where fracture length is much greater than the elevation difference between  
 613 entrances, and  $Pe$  is calculated for  $L = 100h$  in Panel D. Note that this difference has no influence  
 614 on  $Pe$  for the laminar flow regime (i.e. in the left portion of the diagrams). Laminar to turbulent  
 615 transition occurs within the region with the dotted lines.

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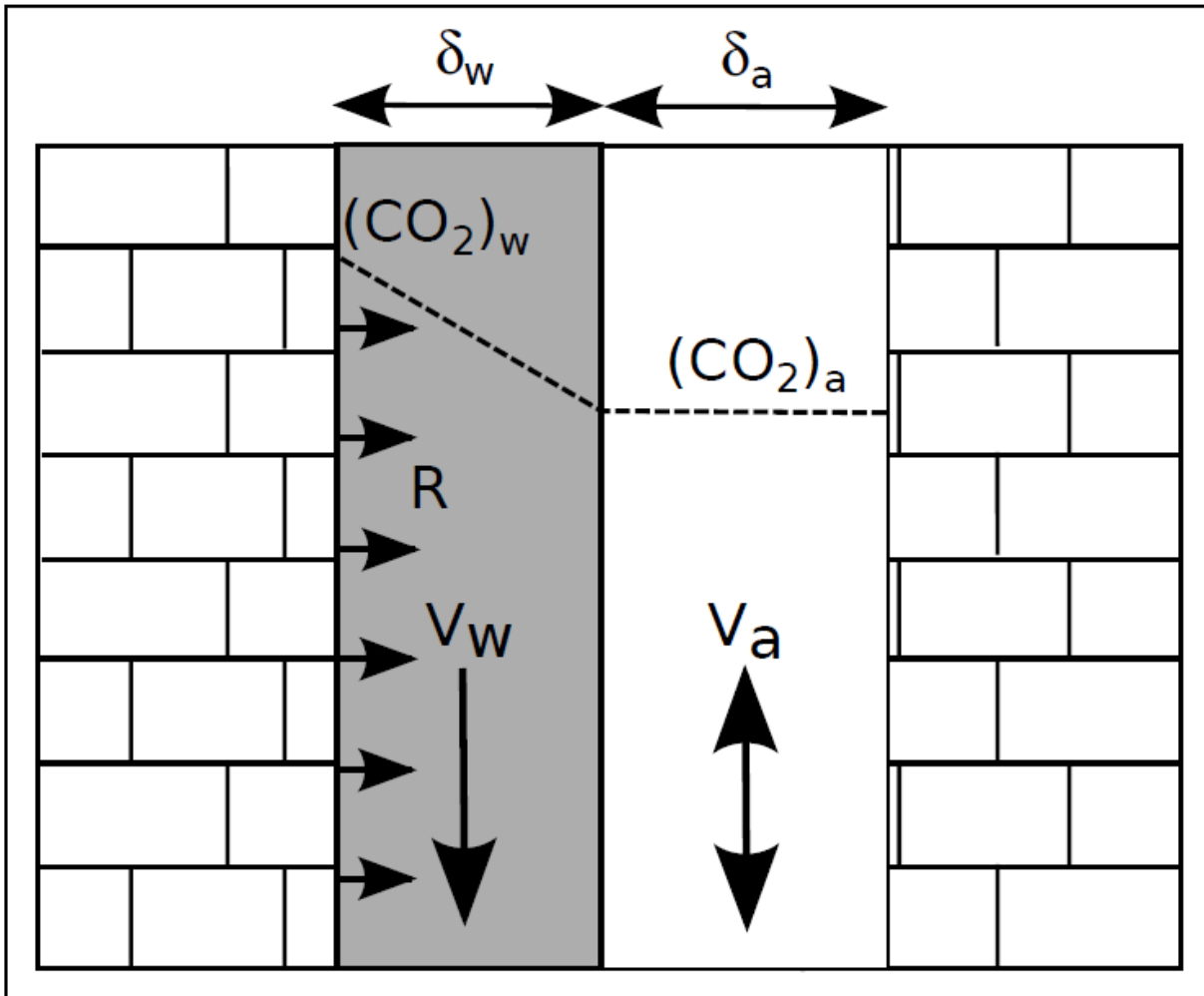


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619 Figure 2. Peclet Numbers (Pe) calculated for chimney effect airflow through fractures with a  
 620 porous cover. In this case, a maximum value of Pe is reached for a sufficiently large fracture  
 621 aperture, as all pressure loss occurs in the porous media in that limit. Panel A depicts a case  
 622 where permeability  $\kappa=10^{-9} \text{ m}^2$ , and Panel B depicts a case where  $\kappa=10^{-12} \text{ m}^2$ . For both cases, the  
 623 cover thickness is assumed to be 5% of the fracture length.

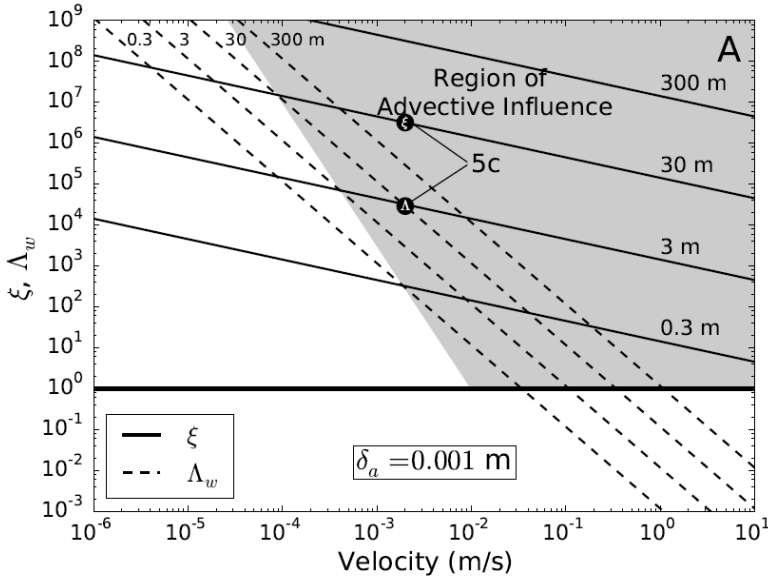


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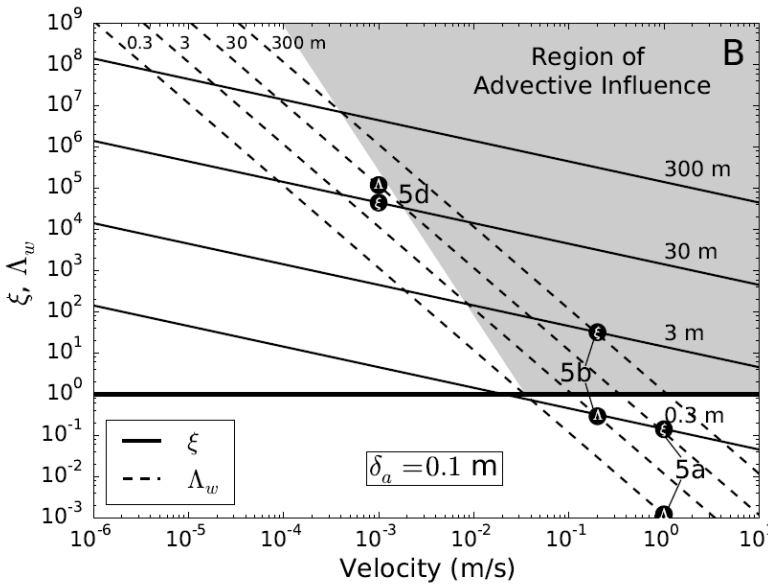
625 Figure 3. The geometry of the assumed model for transport of  $\text{CO}_2$  within the air and water  
 626 flowing in a fracture given by Equations 13 and 14. The dashed line depicts  $\text{CO}_2$  concentration,  
 627 which is assumed to be uniform across the air-filled cross-section and linear across the water  
 628 film. Exchange is assumed to be controlled by the rate of diffusion within the water film.  $R$   
 629 depicts the production of  $\text{CO}_2$  at the rock-water boundary.

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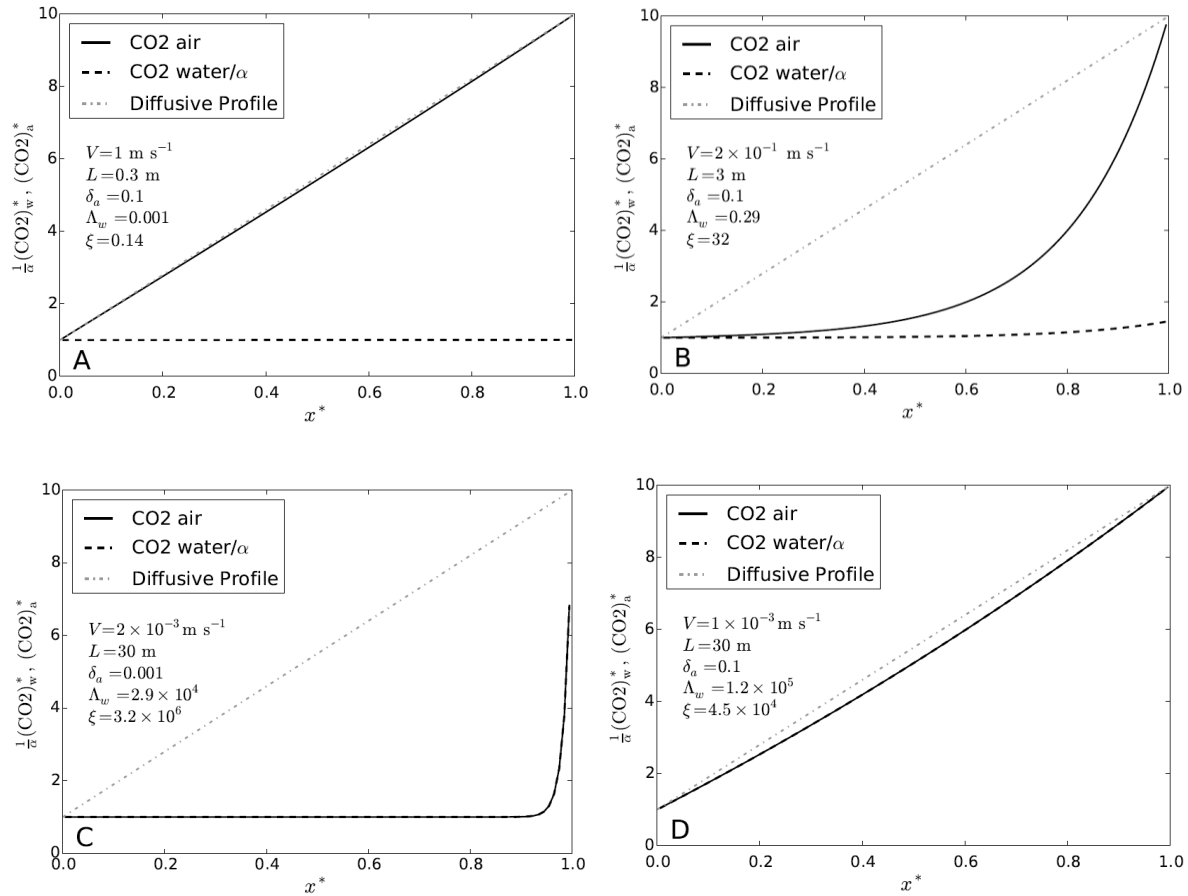


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634 Figure 4. The dimensionless numbers  $\zeta$  (solid) and  $\Lambda_w$  (dashed) as a function of flow velocity for  
 635 different values of fracture length ( $L = 0.3$  m, 3 m, 30 m, and 300 m). Advection is important in  
 636 determining the  $\text{CO}_2$  profile with depth when  $\zeta > 1$  and  $\zeta > \Lambda_w$  (shaded region). Panel (A) shows a  
 637 case where the air-filled fracture aperture is 0.001 m, and Panel (B) shows a case where the air-  
 638 filled fracture aperture is 0.1 m. The letters and numbers (5a-d) label values of  $\zeta$  and  $\Lambda_w$  for the  
 639 example profiles shown in Figure 5.



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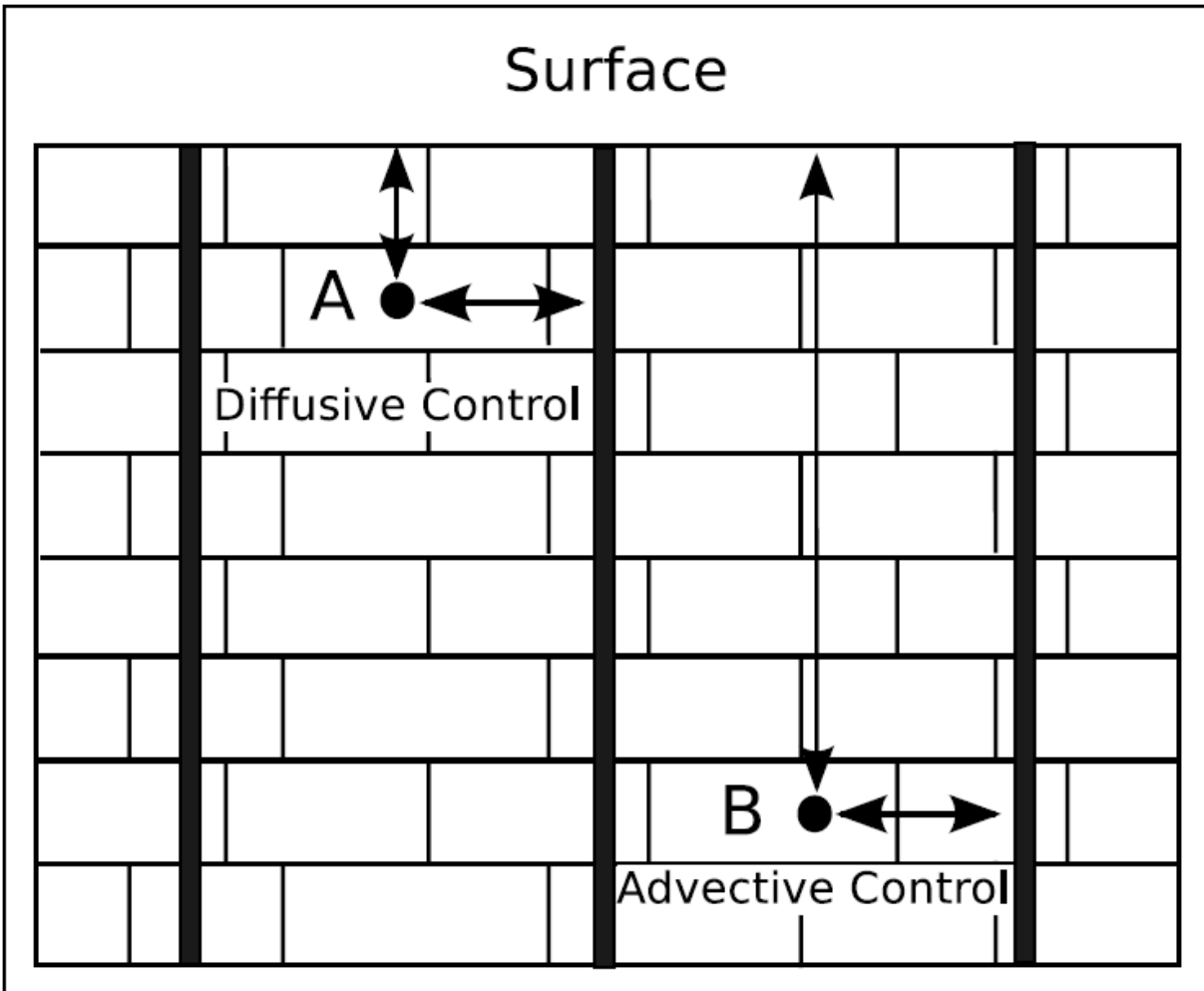
642 Figure 5. Example profiles of water and air CO<sub>2</sub> with depth in the vadose zone for different  
 643 choices of parameters as shown in each panel. These examples assume a source of CO<sub>2</sub> at the  
 644 bottom ( $X^*=1$ ). The solid line depicts the air CO<sub>2</sub> concentration, the black dashed line depicts the  
 645 water CO<sub>2</sub> concentration, and the grey dashed line depicts the purely diffusive profile for air  
 646 CO<sub>2</sub>. The location of each case within the parameter space is also depicted in Figure 4. The water  
 647 film influences the CO<sub>2</sub> air concentration when  $\xi > 1$  and  $\xi > \Lambda_w$ . When  $\Lambda_w > 1$  then the CO<sub>2</sub> profile  
 648 in the water is closely coupled to the CO<sub>2</sub> profile in the air.

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654 Figure 6. A profile through the vadose zone with three advectively dominated vertical fractures  
 655 through the rock. At point A, dynamics will be largely controlled by diffusion, because it is as  
 656 close to the surface as it is to the fractures. However, at point B advection is likely to influence  
 657 CO<sub>2</sub> concentration, since diffusion to the advective fracture will be much faster than diffusion to  
 658 the surface. Though much of the vadose zone may not be in close proximity to a fracture flow  
 659 path dominated by advection, many points within the vadose zone will be closer to such a path  
 660 than they are to the surface.